

Atomistic Simulation in the Model HEA: CuNiCoFe

D. Utt, A. Stukowski, K. Albe



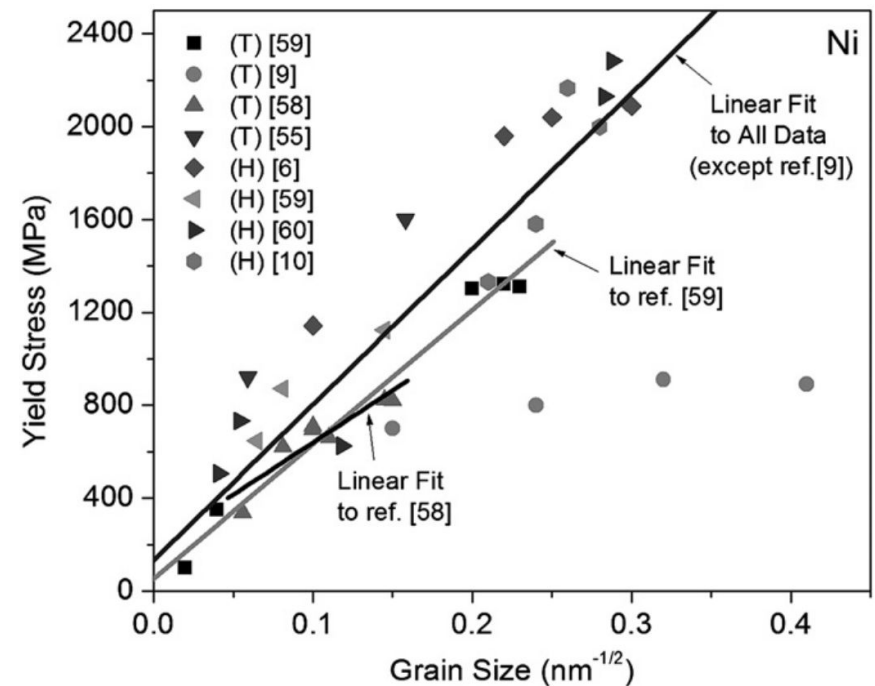
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DFG Deutsche
Forschungsgemeinschaft

Motivation

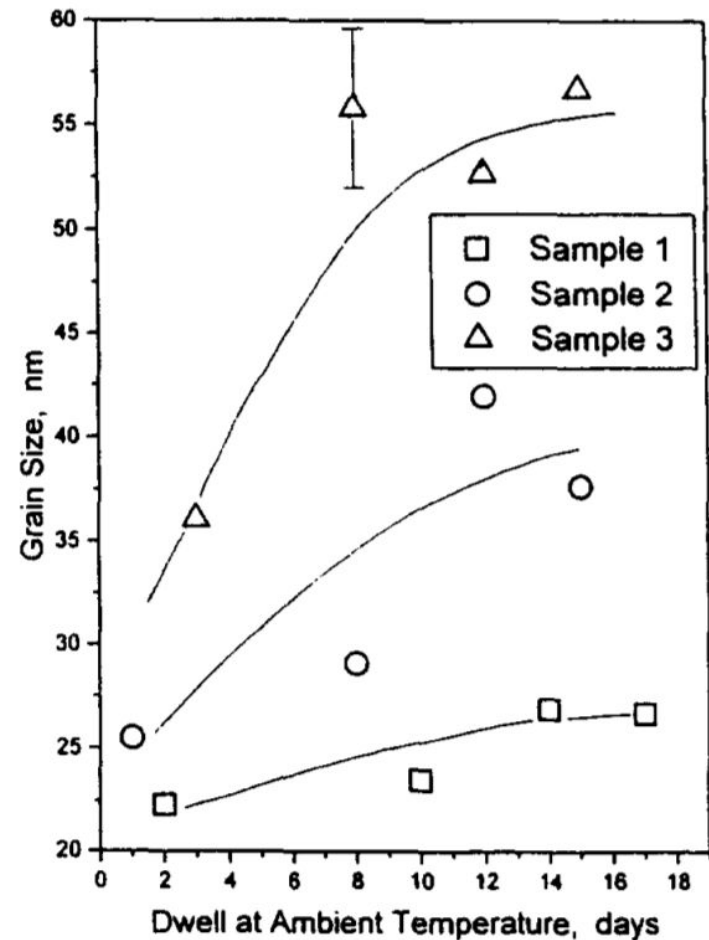
- Nanocrystalline metals show outstanding mechanical properties.



Carlton and Ferreira, Acta Materialia 55 (11), 2007

Motivation

- Nanocrystalline metals show outstanding mechanical properties.
- Small grain sizes are not stable in pure materials.



Gertsman and Birringer, Scripta Metallurgica et Materialia 30 (5), 1994

Motivation



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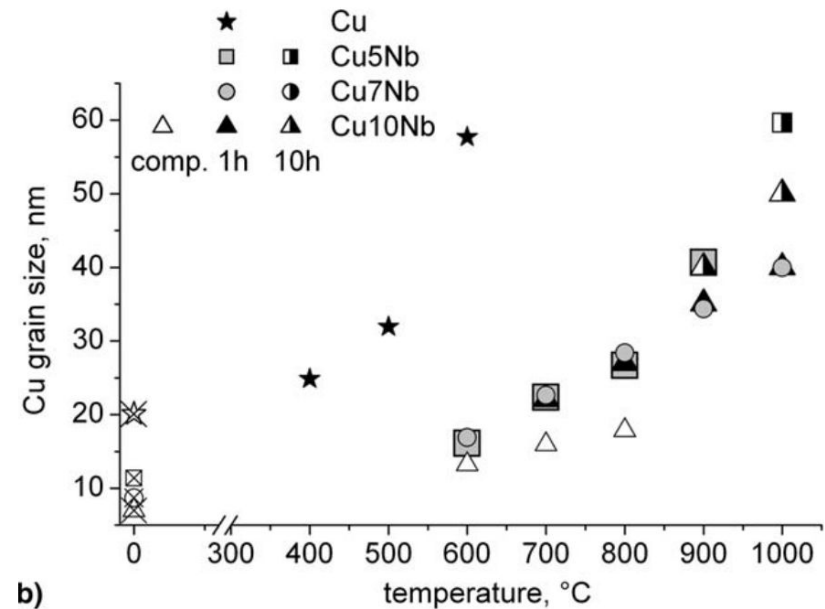
- Secondary elements can be added to stabilize the microstructure.

Motivation

- Secondary elements can be added to stabilize the microstructure.

1. Kinetically:

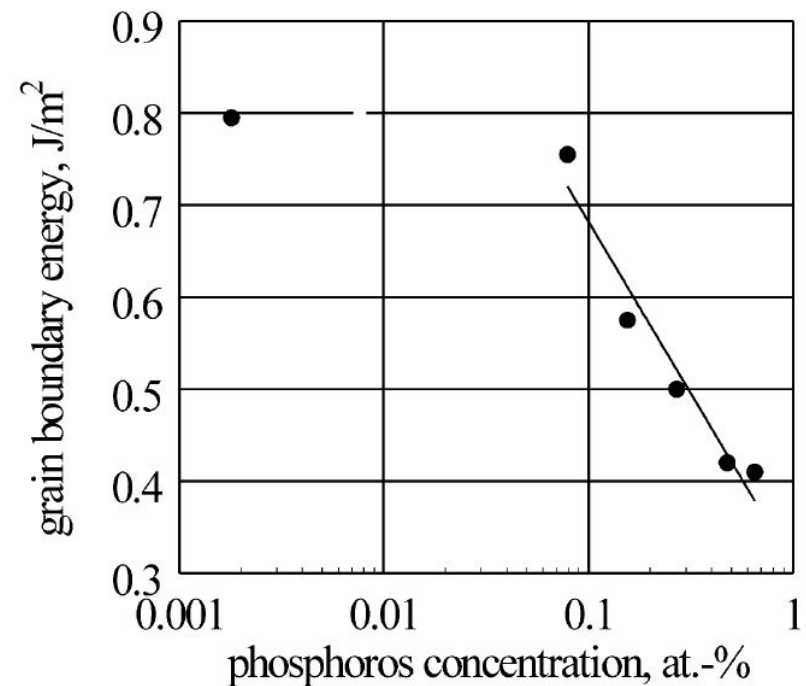
- Solute drag mechanism
- Zener pinning



Cahn, Acta Metallurgica 10 (9), 1962
Nes et al., Acta Metallurgica 33 (1), 1985
Botcharova et al., Acta Materialia 51 (12), 2006

Motivation

- Secondary elements can be added to stabilize the microstructure.
1. Kinetically
 2. Thermodynamically:
 - Segregation reduces GB energy to 0.
 - Removes driving force for grain growth.



Weissmüller, Nanostructured Materials, 3 (1-6), 1993
Kirchheim, Acta Materialia, 50 (2), 2002

Motivation



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- Slower grain growth in HEAs?

Motivation

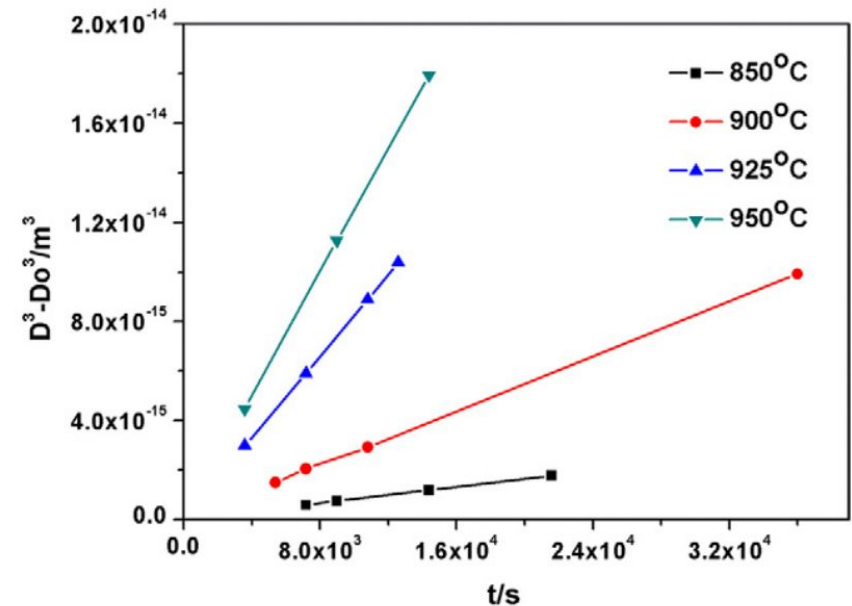
- Slower grain growth in HEAs?
 - Local disorder



- Slower grain growth in HEAs?
 - Local disorder
 - Segregation effects

Motivation

- Slower grain growth in HEAs?
 - Local disorder
 - Segregation effects
- Observed in microcrystalline FeCoNiCrMn
 - growth exponent = 3



Liu et al., Scripta Materialia 68 (7), 2013

What is the origin of the reduced grain growth in HEAs?

Understanding kinetic and thermodynamic aspects.

Molecular Dynamics Study

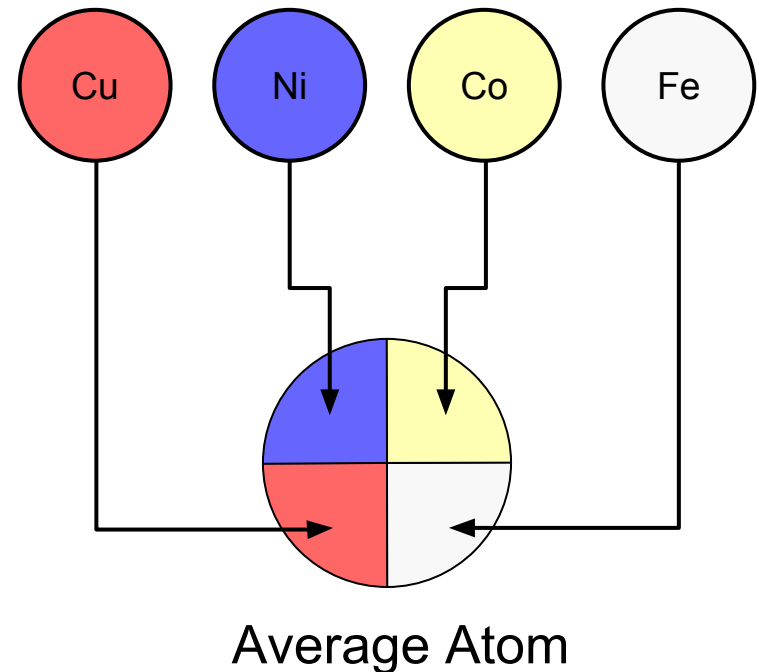


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- Compare grain growth:
 - Cu
 - CuNiCoFe
 - Average atom

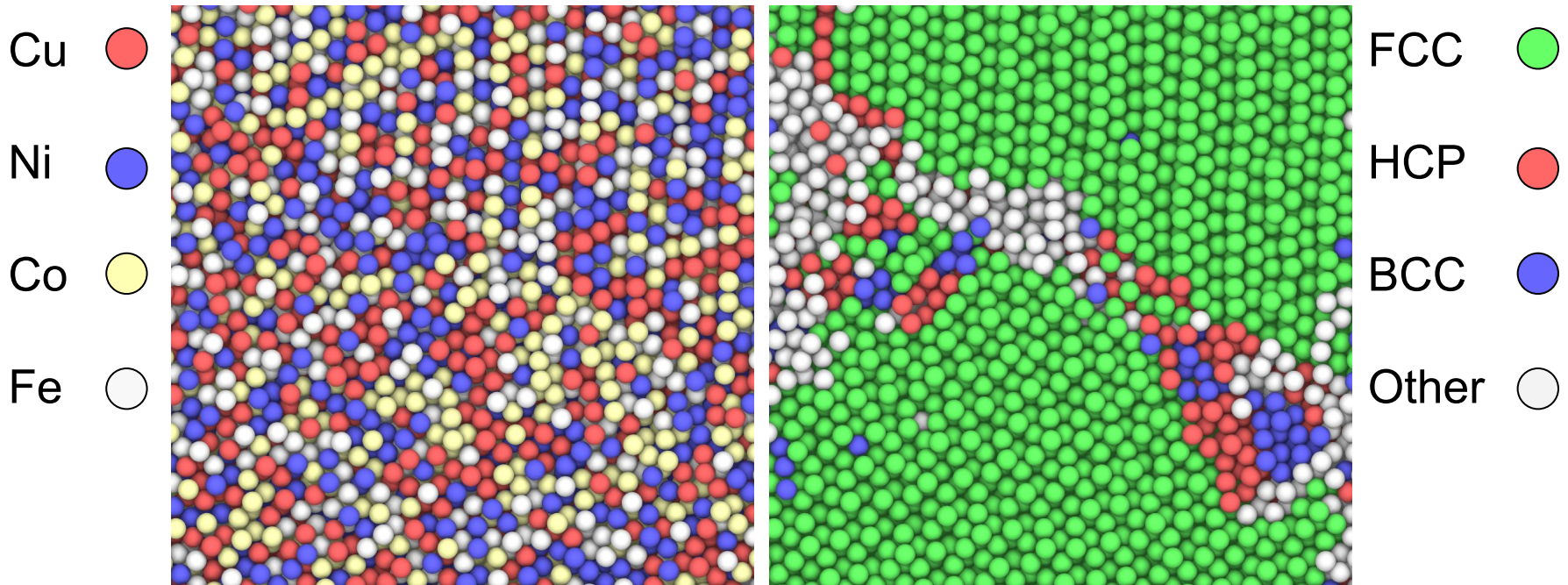
Molecular Dynamics Study

- Compare grain growth
- Average atom
 - Artificial element
 - Behaves like CuNiCoFe alloy
 - Leads to average matrix without local disorder
 - Selectively study effects of local disorder



Molecular Dynamics Study

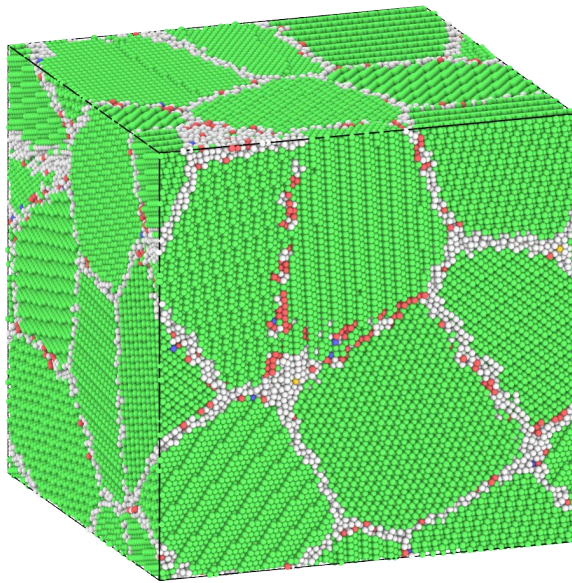
- Compare grain growth
- Average atom
- Atoms will be color coded



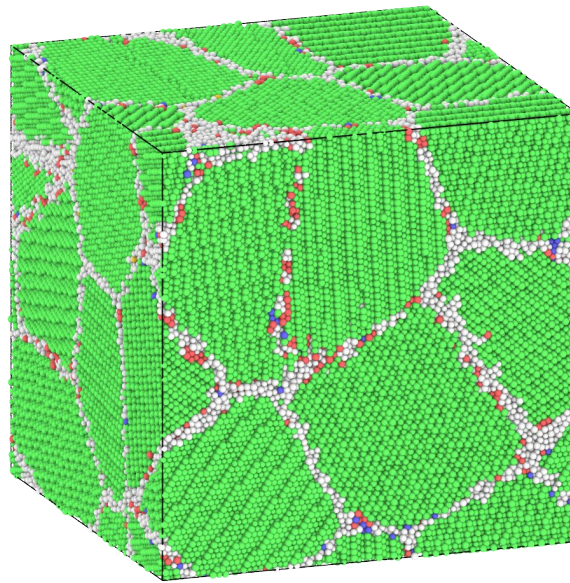
Stukowski, Modelling and Simulation in Materials Science and Engineering 18 (1) 2009

$$t = t_{\text{initial}}$$

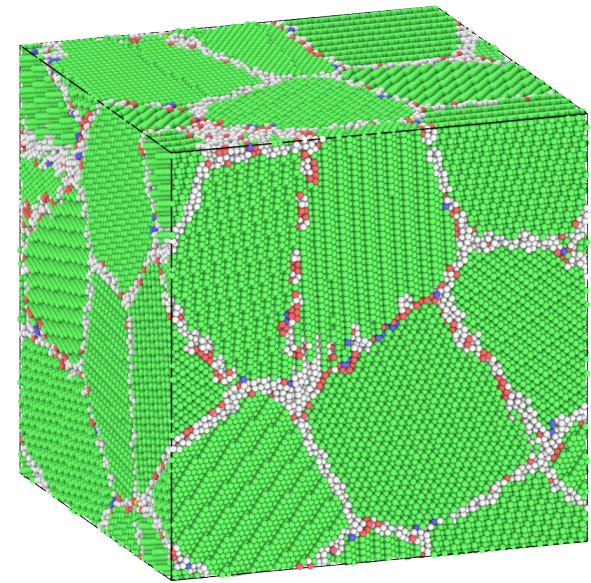
Cu



CuNiCoFe



Average Atom

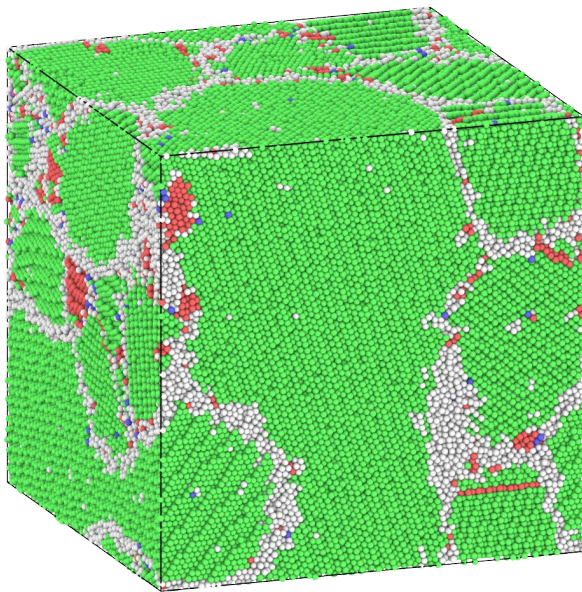


FCC ● HCP ● BCC ● Other ○

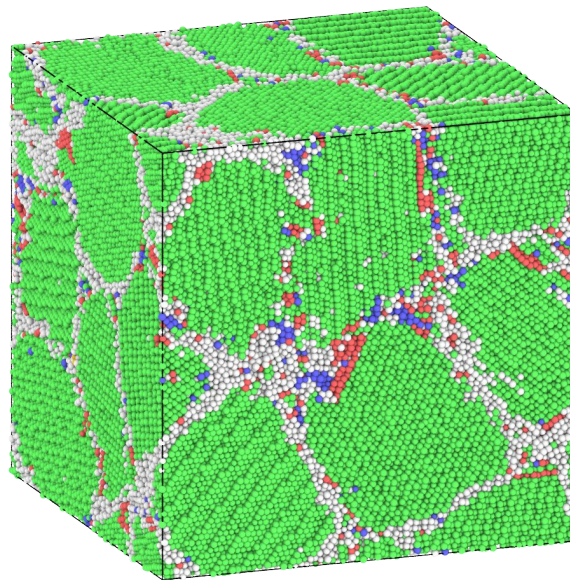
$$t = 0.33 t_{\text{final}}$$



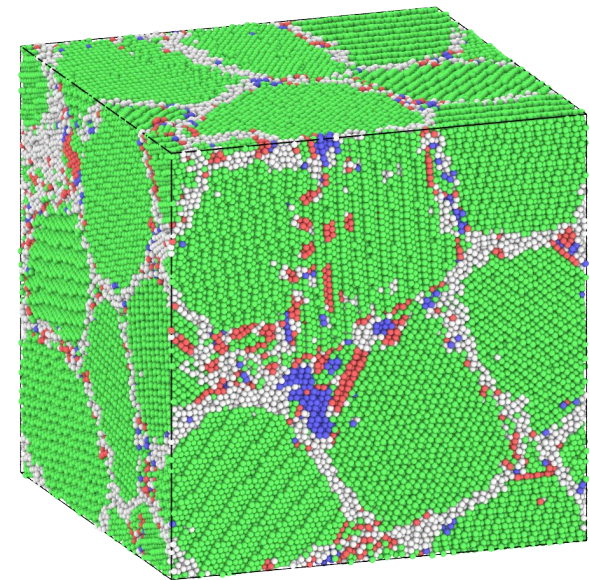
Cu



CuNiCoFe



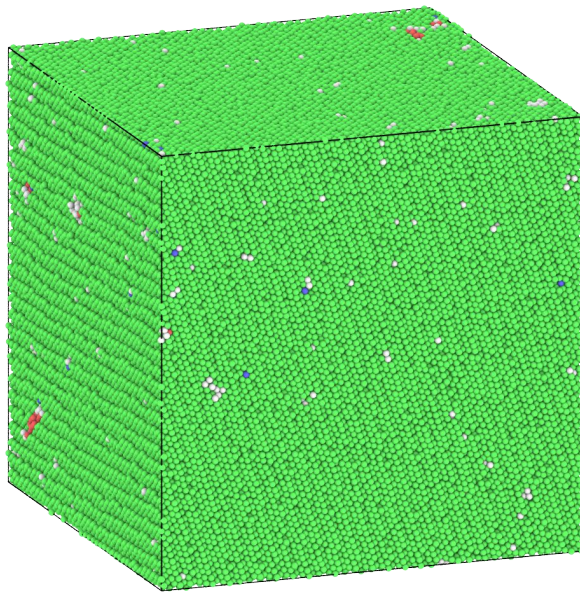
Average Atom



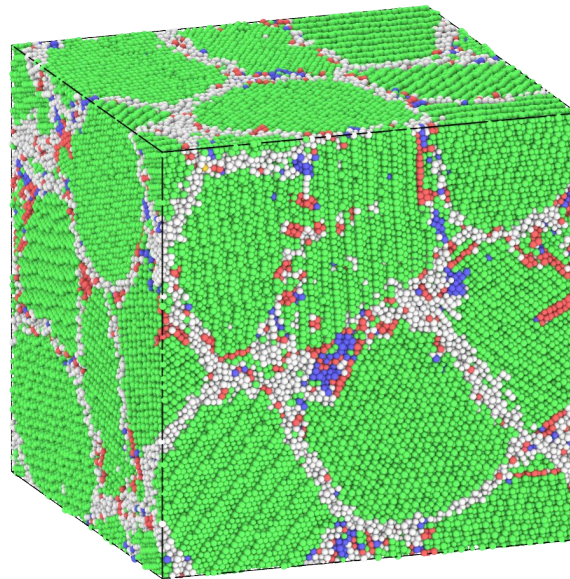
FCC ● HCP ● BCC ● Other ○

$$t = t_{\text{final}}$$

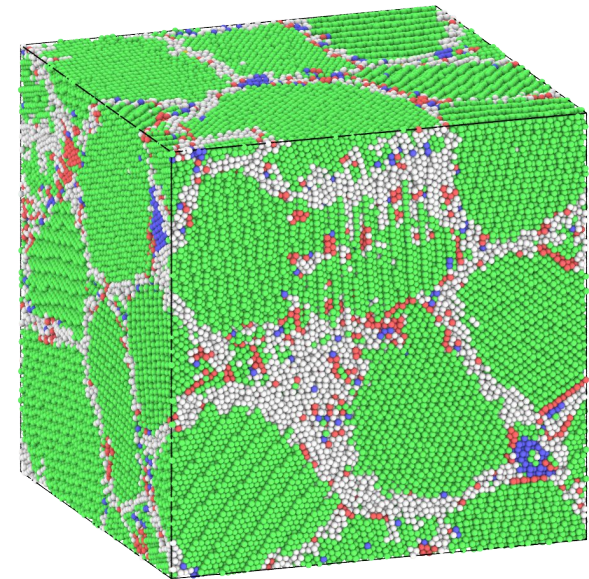
Cu



CuNiCoFe



Average Atom



FCC ● HCP ● BCC ● Other ○

$$t = t_{\text{final}}$$

Cu

CuNiCoFe

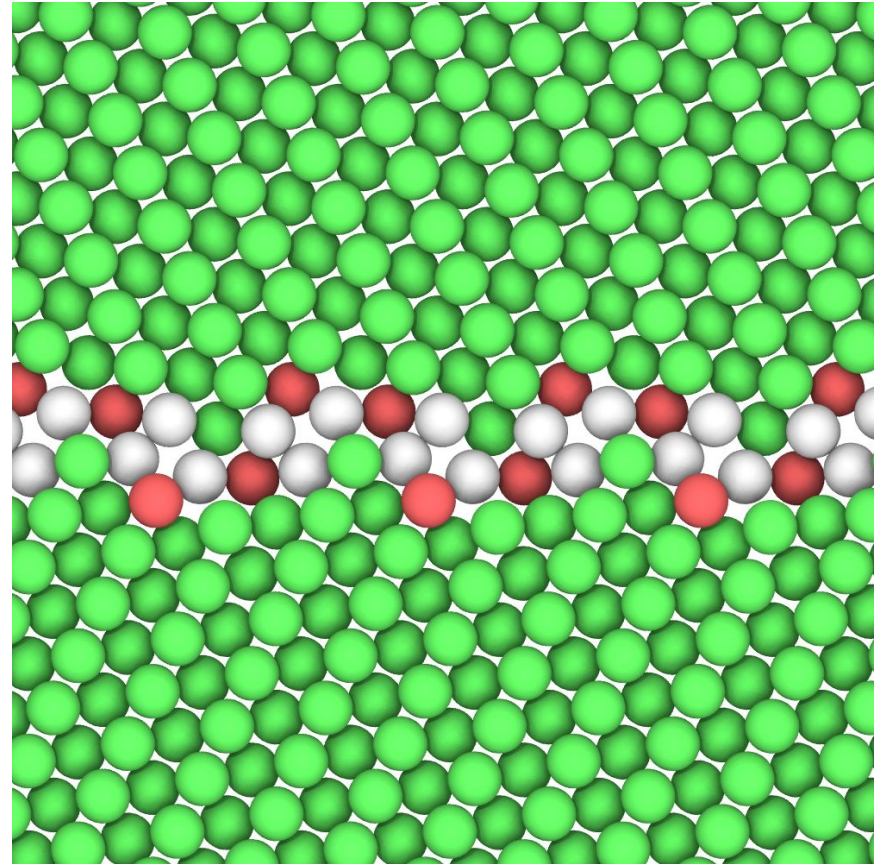
Average Atom

Local disorder is not the main contributor to the reduced grain growth.

FCC ● HCP ● BCC ● Other ○

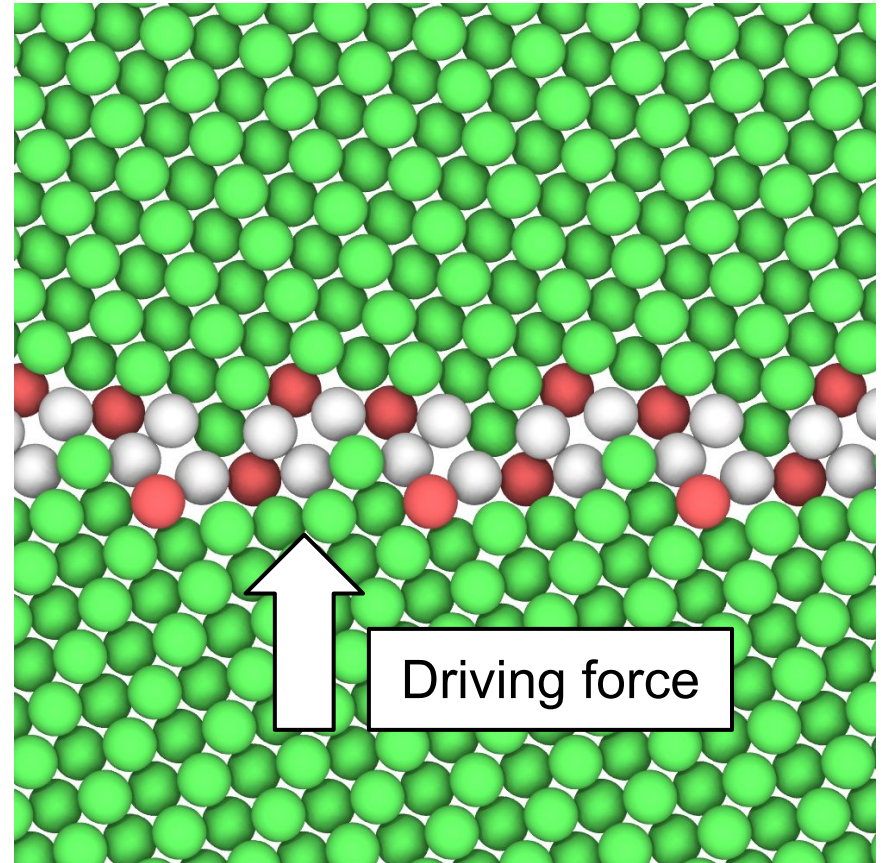
GB Mobility

- $\Sigma=11$ (332) symmetrical tilt GB
- Bicrystalline simulation setup



GB Mobility

- $\Sigma=11$ (332) symmetrical tilt GB
- Bicrystalline simulation setup
- Apply artificial driving force to GB

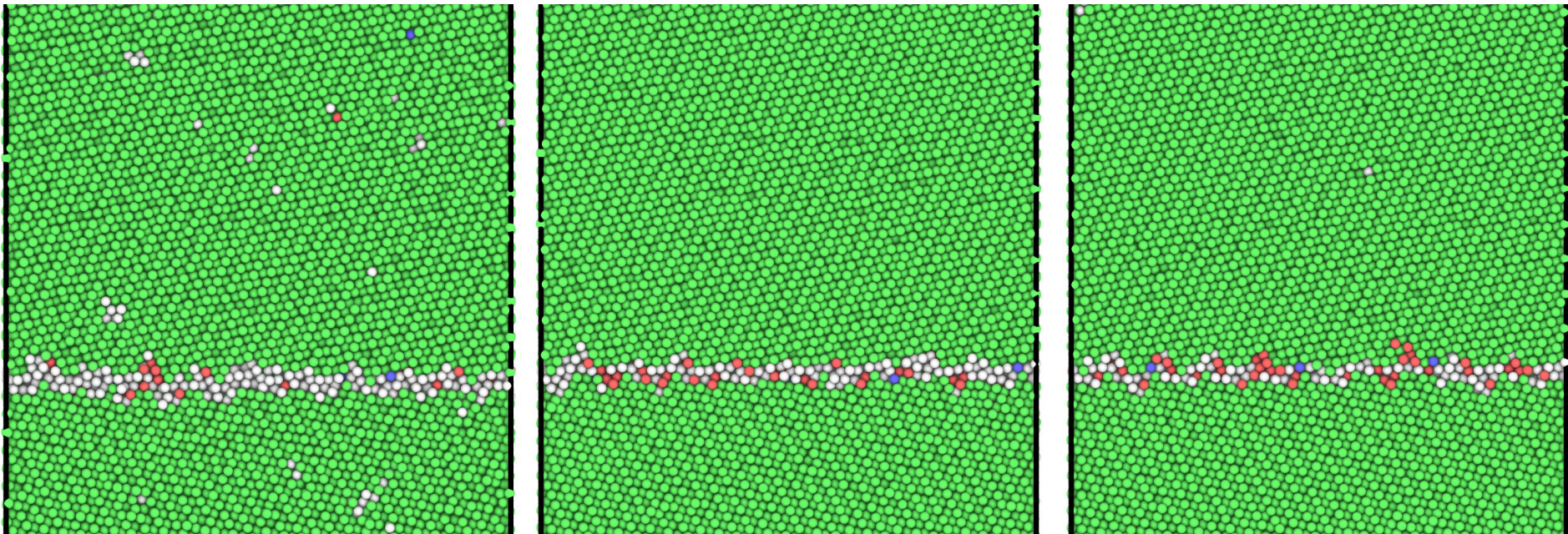


$$t = t_{\text{initial}}$$

Cu

CuNiCoFe

Average Atom



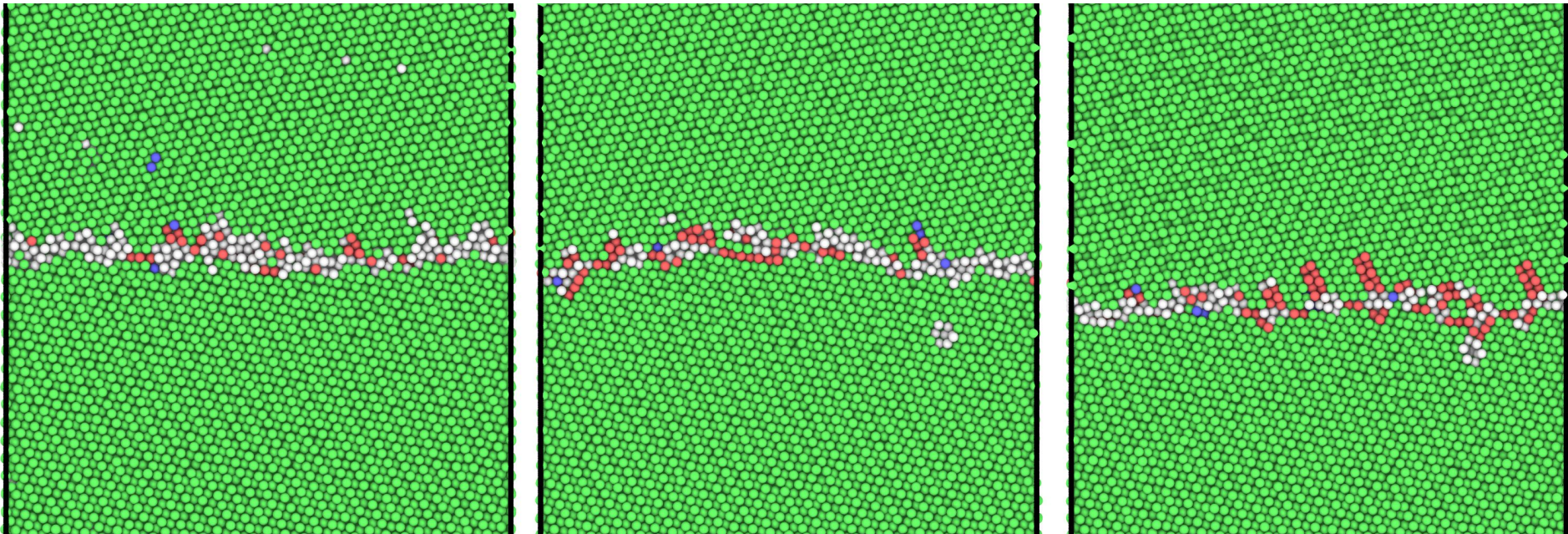
FCC ● HCP ● BCC ● Other ○

$$t = 0.5 t_{\text{final}}$$

Cu

CuNiCoFe

Average Atom



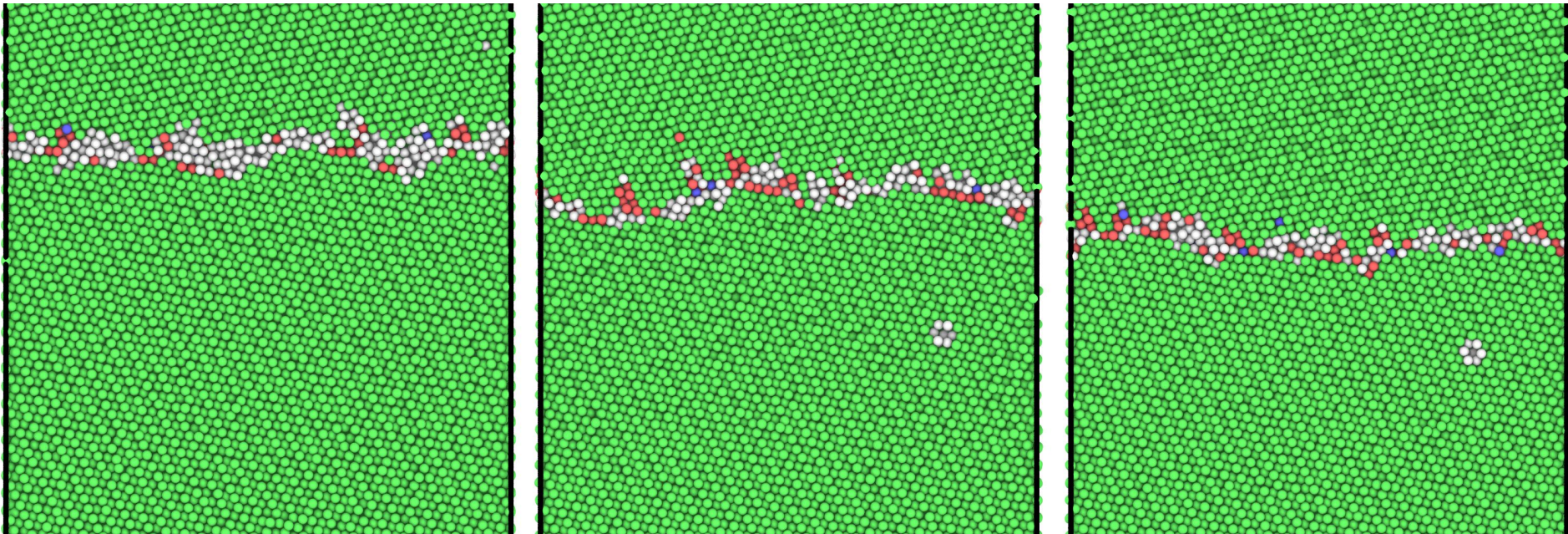
FCC ● HCP ● BCC ● Other ○

$$t = t_{\text{final}}$$

Cu

CuNiCoFe

Average Atom



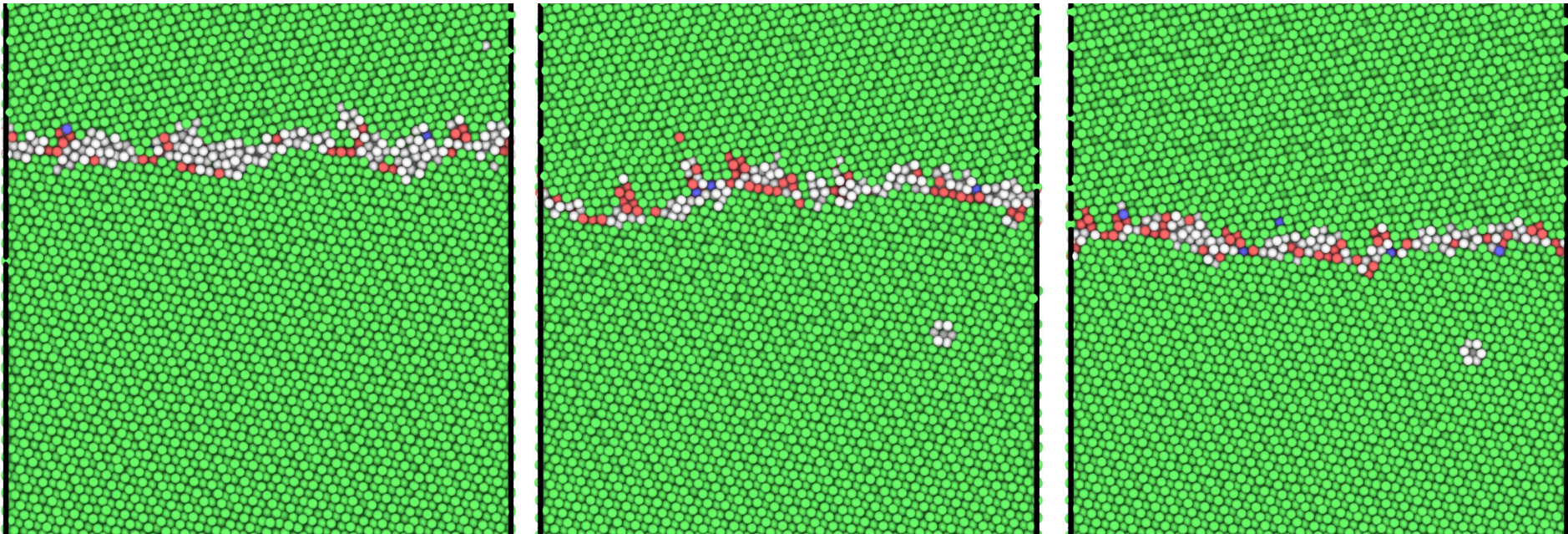
FCC ● HCP ● BCC ● Other ○

$$t = t_{\text{final}}$$

Cu

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Average Atom



FCC ● HCP ● BCC ● Other ○

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Cu

CuNiCoFe

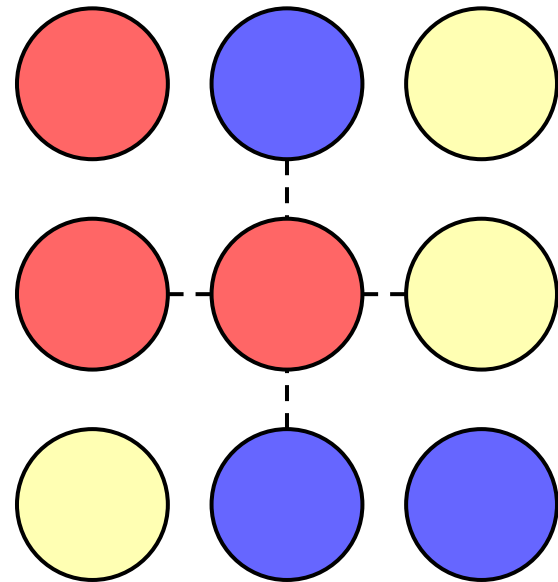
Average Atom

No signs of pinning in the CuNiCoFe.
No evidence for solute drag.

FCC ● HCP ● BCC ● Other ○

Simplified Lattice Monte-Carlo Model

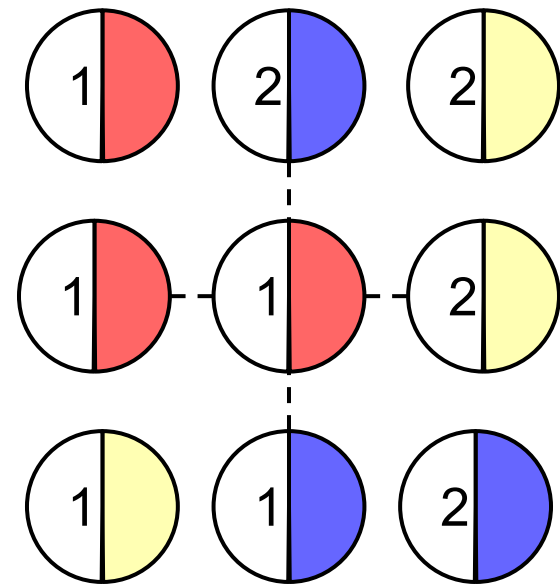
- Fixed FCC lattice considering only nearest neighbor interactions



Chookajorn and Schuh, Physical Review B 89 (6), 2014

Simplified Lattice Monte-Carlo Model

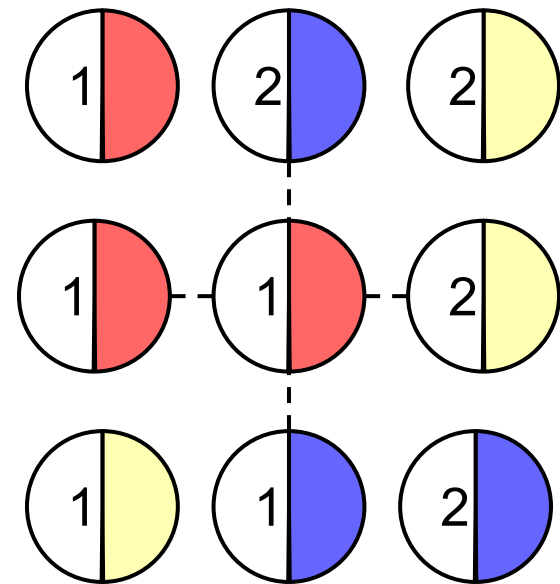
- Fixed FCC lattice considering only nearest neighbor interactions
- Every atom has two properties
 - Atom type (color)
 - Grain number (number)



Chookajorn and Schuh, Physical Review B 89 (6), 2014

Simplified Lattice Monte-Carlo Model

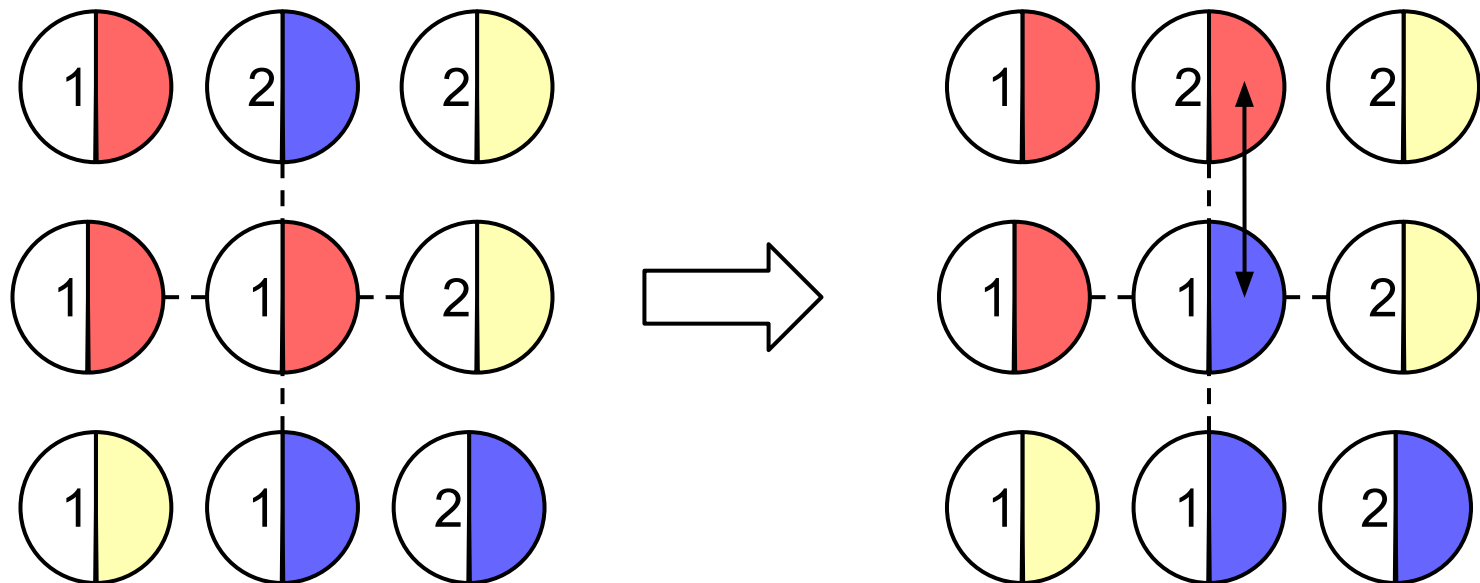
- Fixed FCC lattice considering only nearest neighbor interactions
- Every atom has two properties
 - Atom type (color)
 - Grain number (number)
- Periodic boundary conditions



Chookajorn and Schuh, Physical Review B 89 (6), 2014

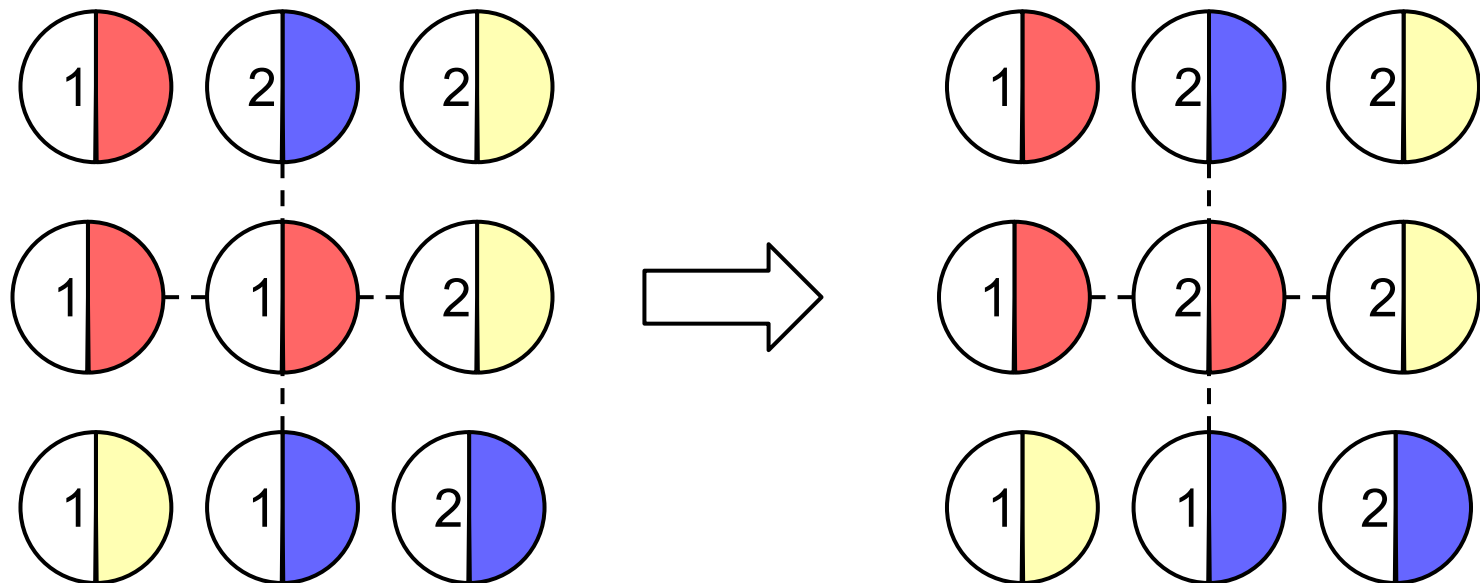
Simplified Lattice Monte-Carlo Model

- Each step an atom can swap:
 - Atom type



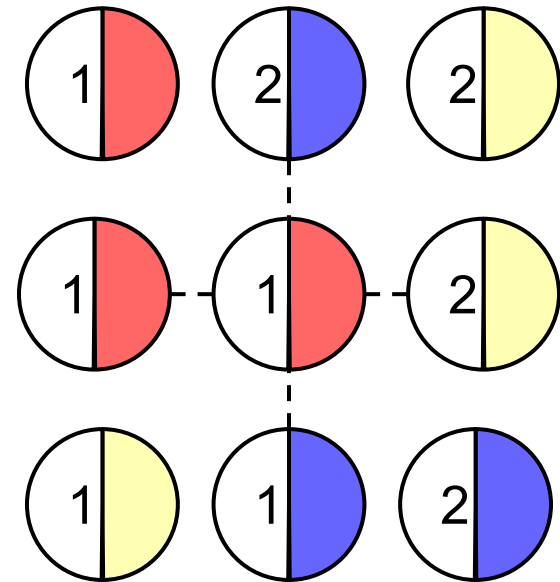
Simplified Lattice Monte-Carlo Model

- Each step an atom can swap:
 - Atom type
- Each step an atom change:
 - Grain number



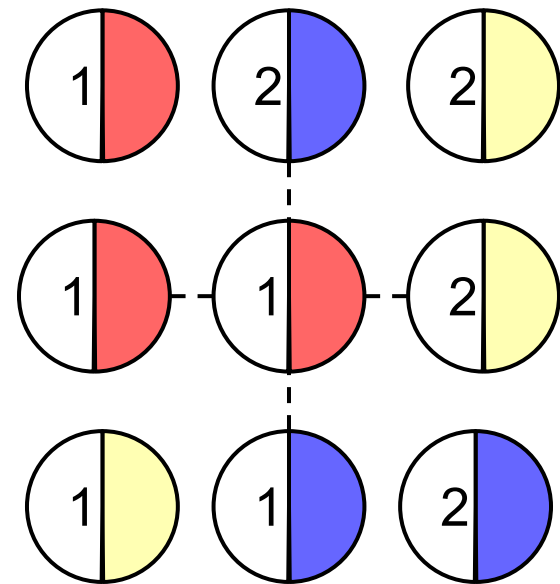
Simplified Lattice Monte-Carlo Model

- Site energy depends on direct neighbors.



Simplified Lattice Monte-Carlo Model

- Bond energy varies with direct neighbors.
- Parameters space:
 - Binary mixing enthalpies
 - Binary segregation enthalpies
 - Unary GB energies (γ_0)



Simplified Lattice Monte-Carlo Model

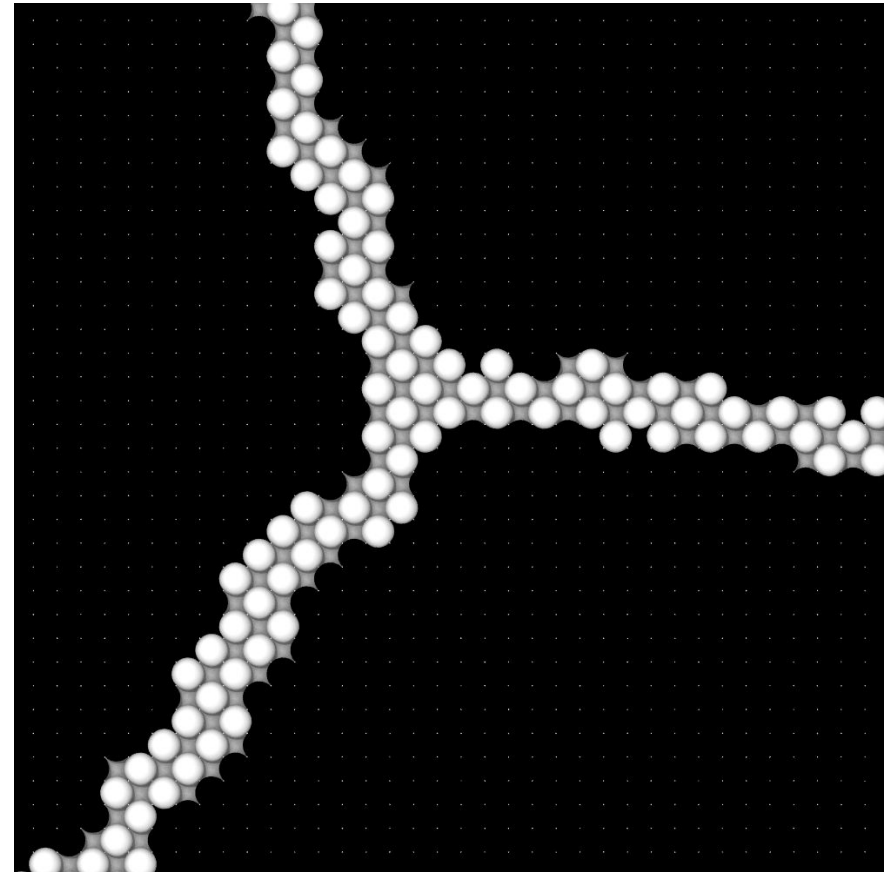


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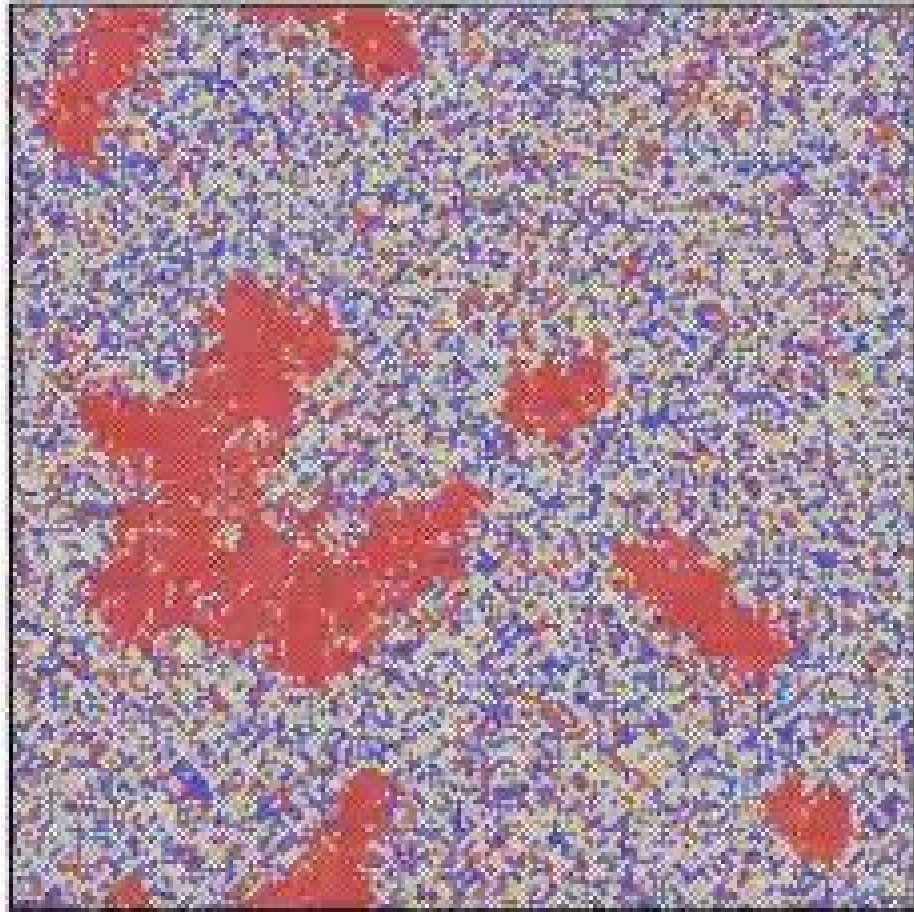
- Set all parameters
- Varied only GB energies (γ)
- Simulated annealing

Simplified Lattice Monte-Carlo Model

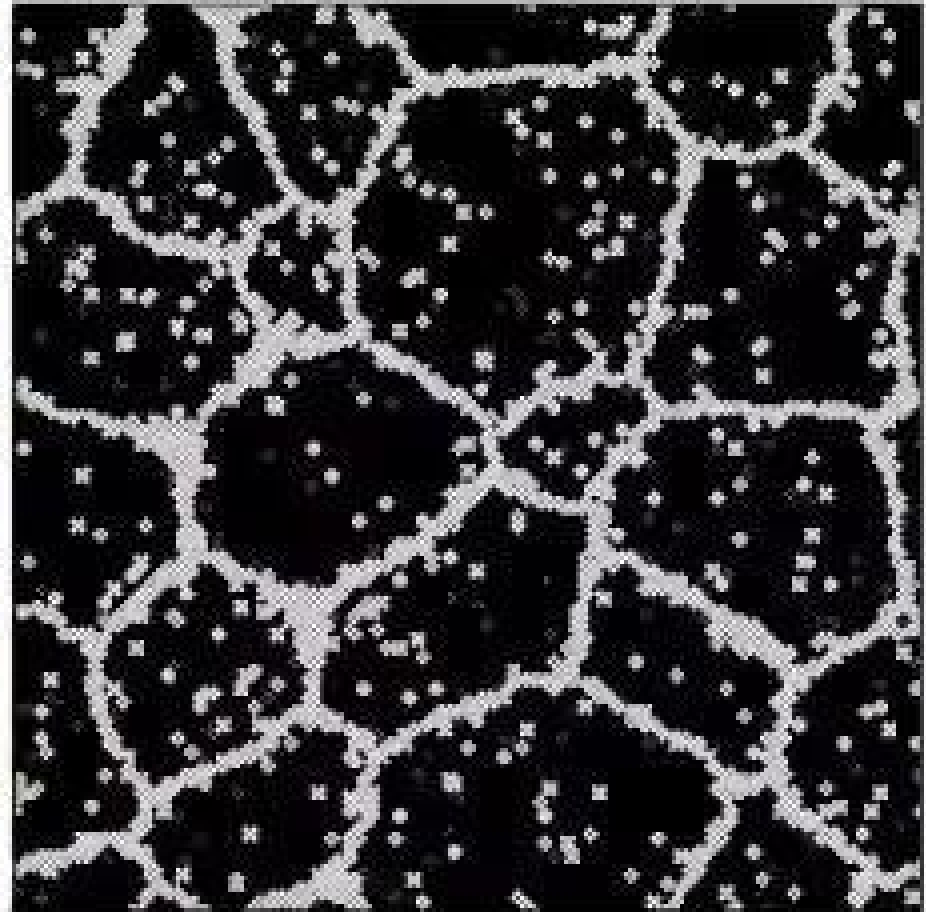
- Set all parameters
- Varied only GB energies (γ)
- Simulated annealing
- Atoms will be color coded
 - Type
 - Grain boundary
 - On boundary
 - Grain interior



$$\gamma = 5.5\gamma_0$$

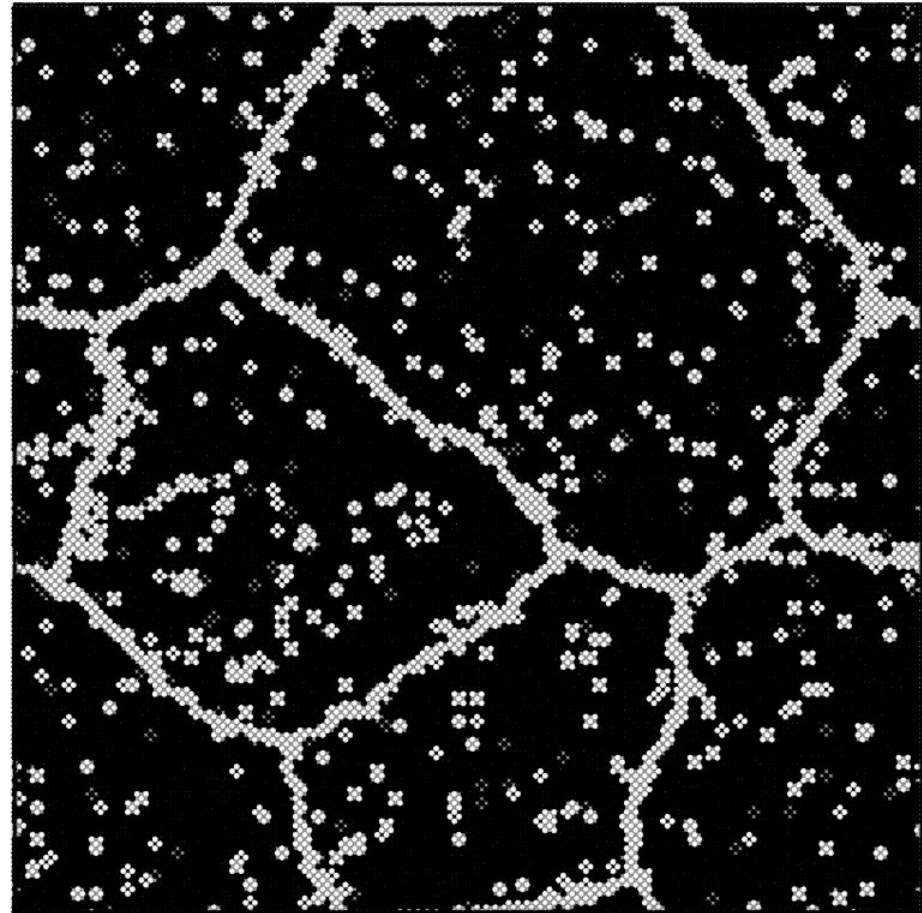
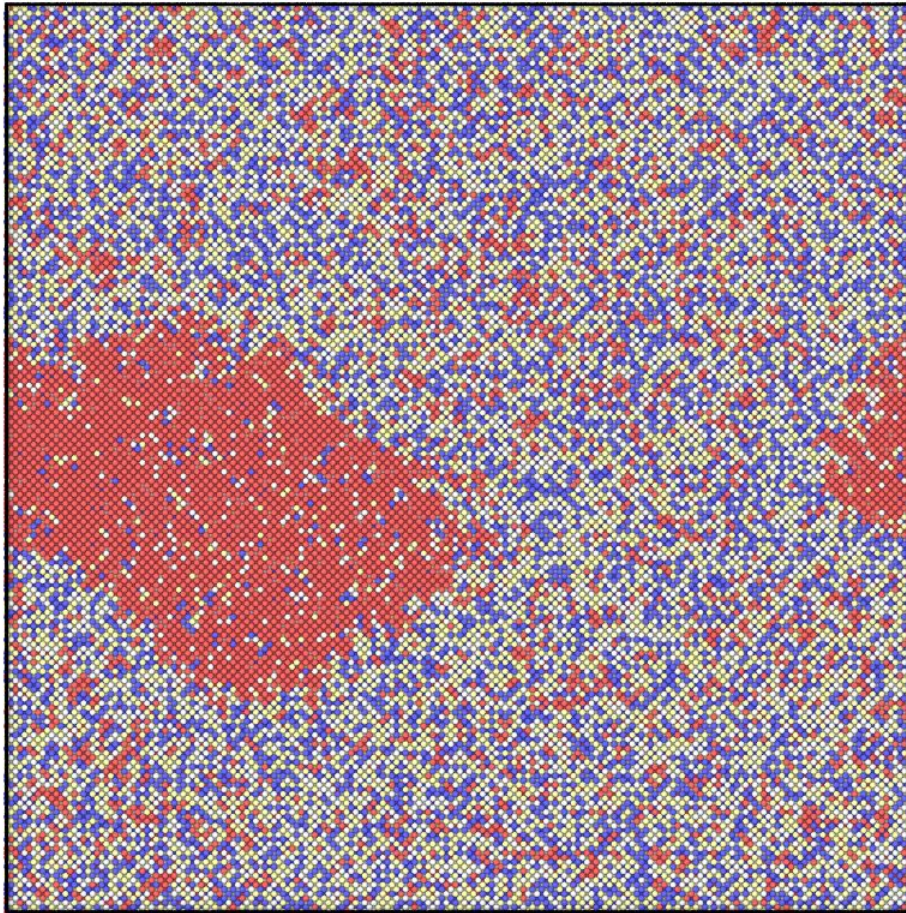


Atom Types

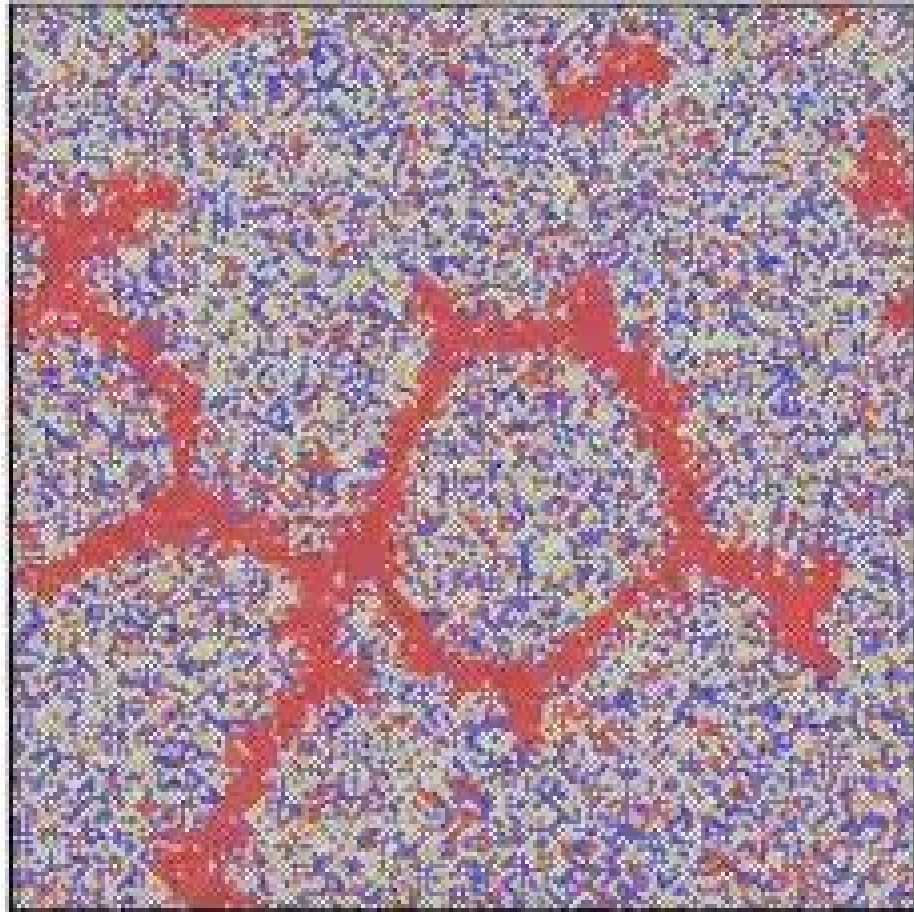


Grain Boundaries

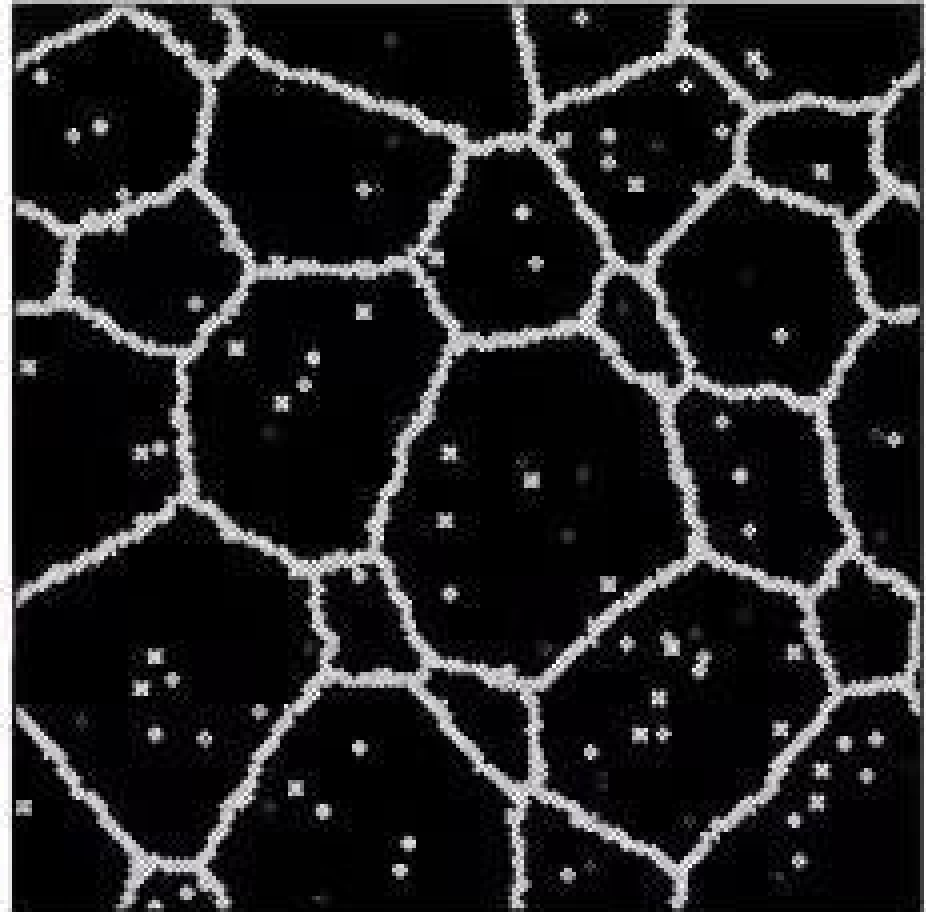
$\gamma = 5.5 \gamma_0$ Final Configuration



$$\gamma = 8\gamma_0$$



Atom Types

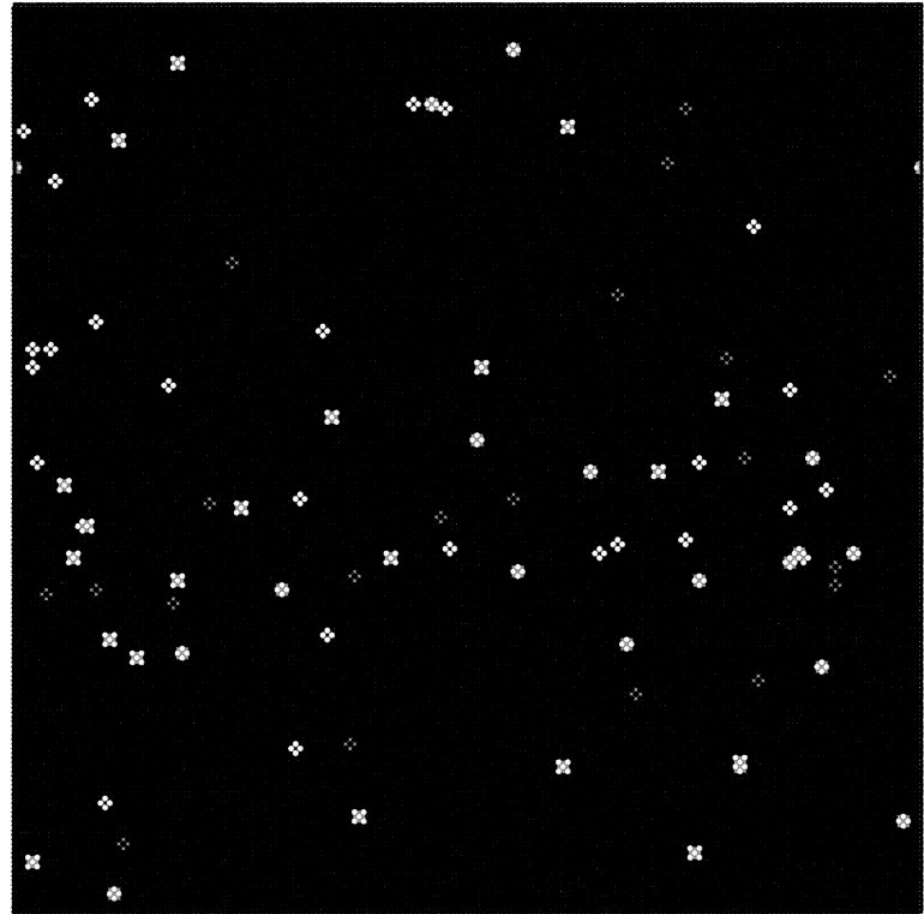
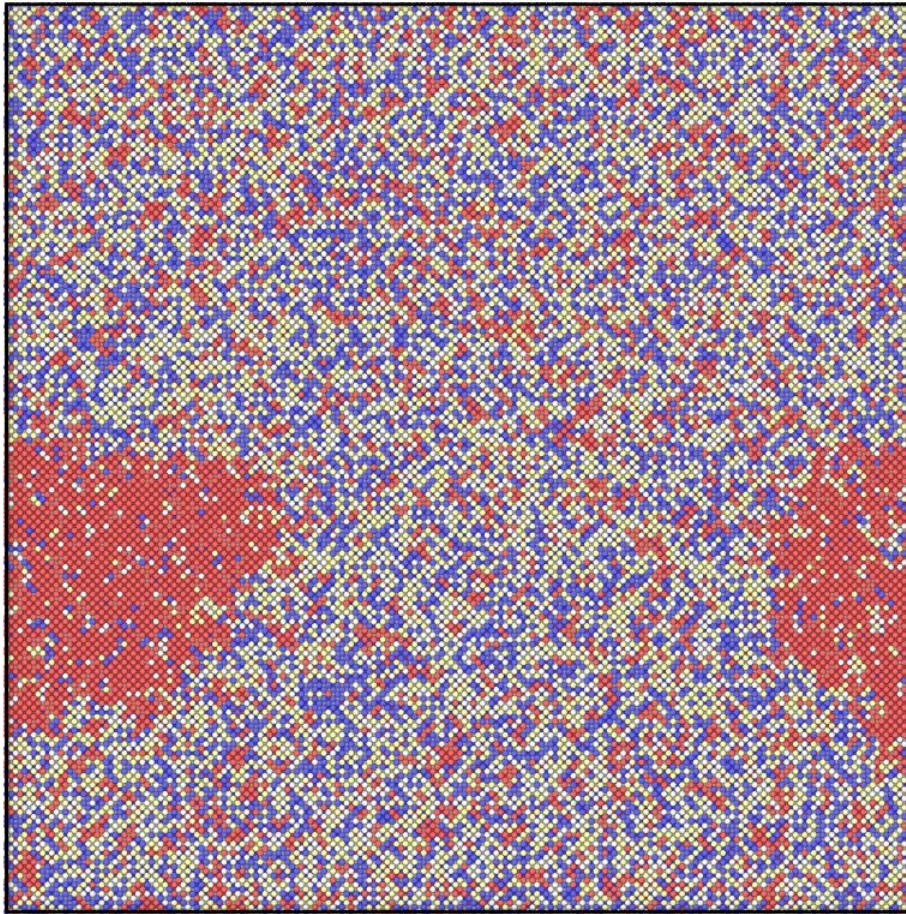


Grain Boundaries

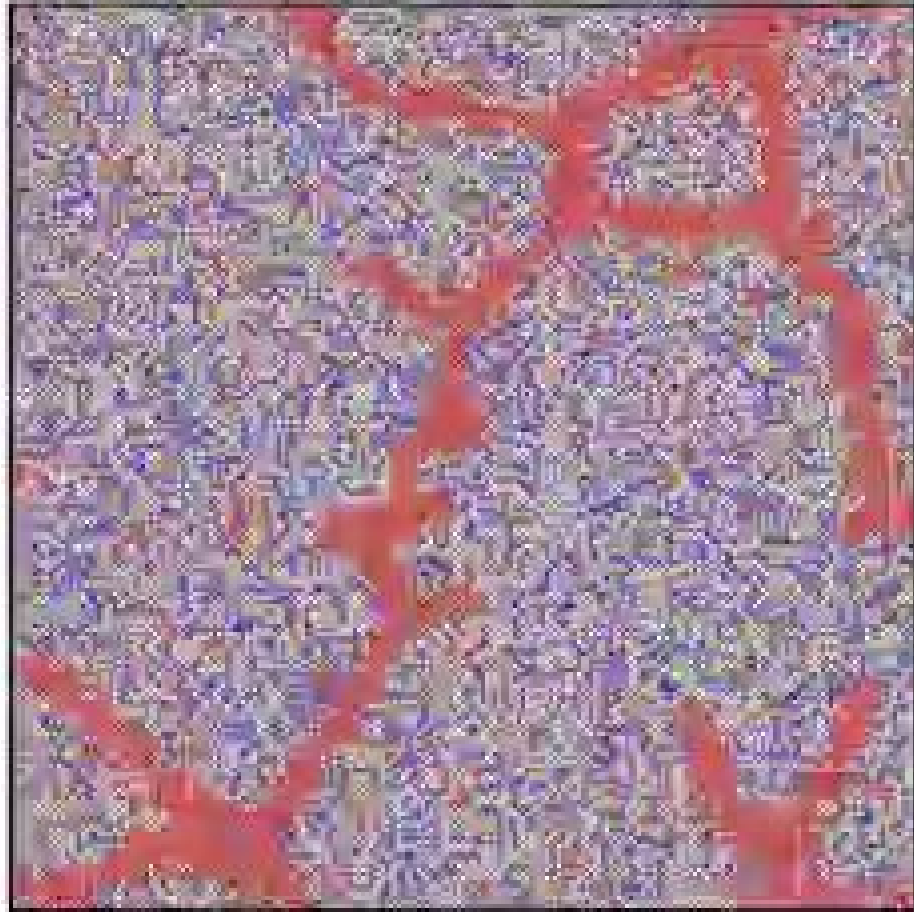
$\gamma = 8 \gamma_0$ Final Configuration



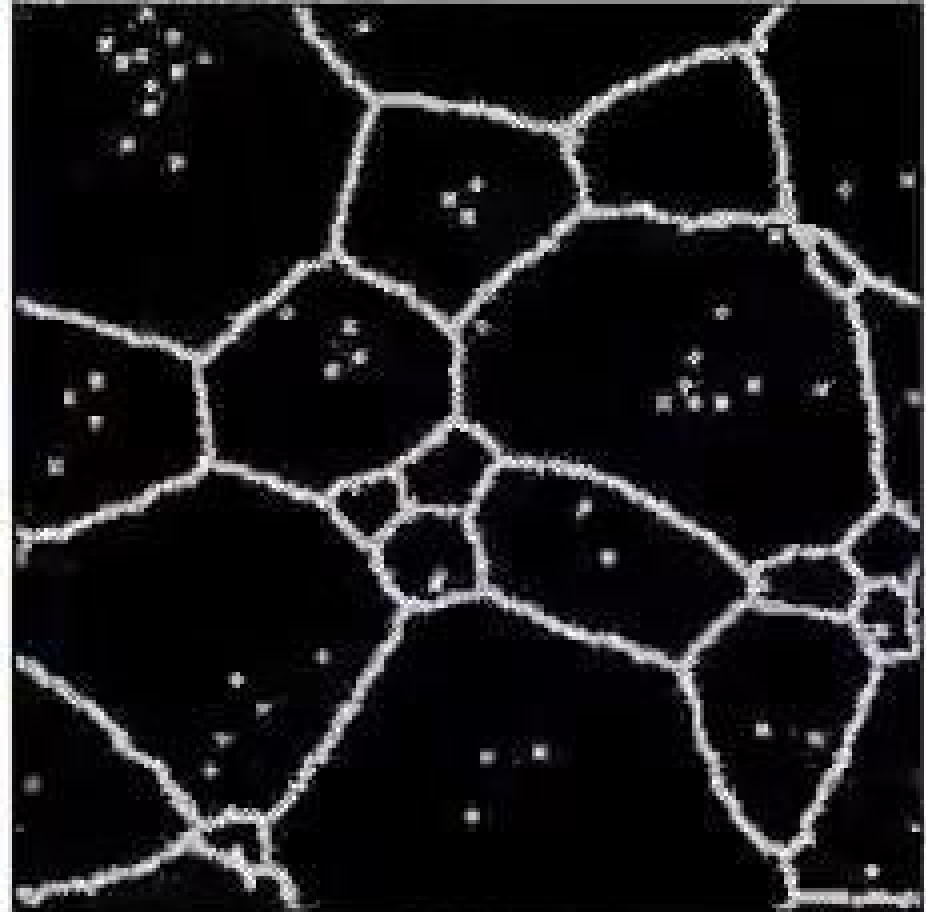
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$$\gamma = 9.5\gamma_0$$



Atom Types

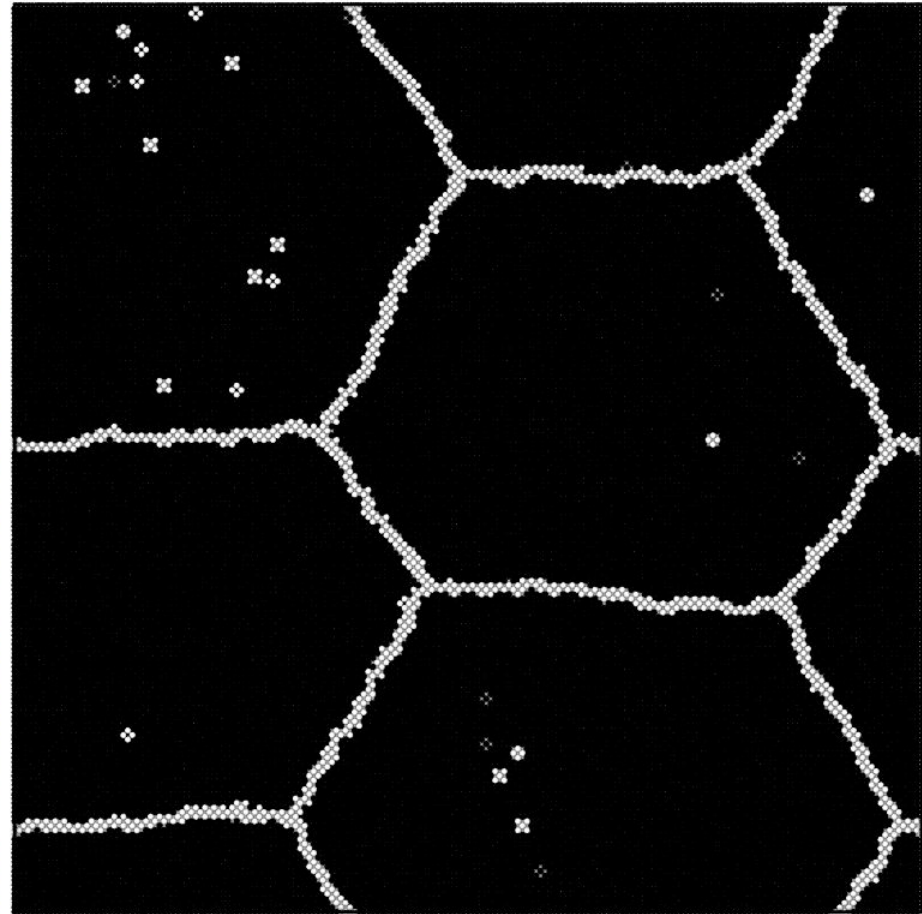
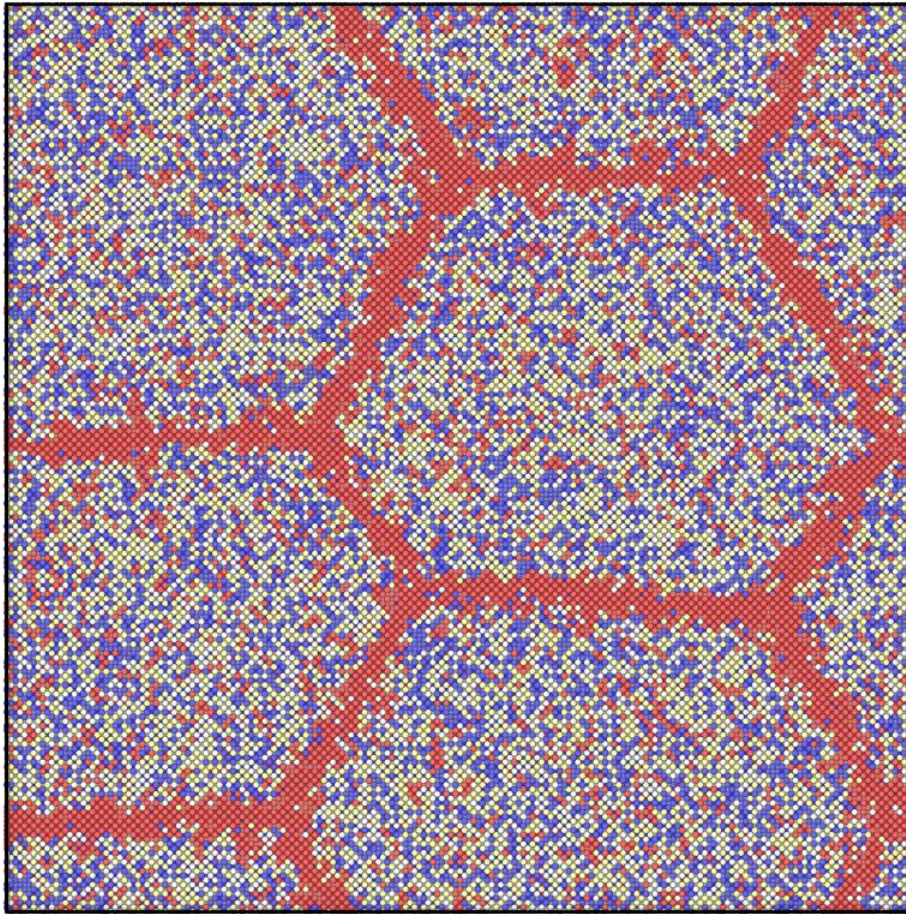


Grain Boundaries

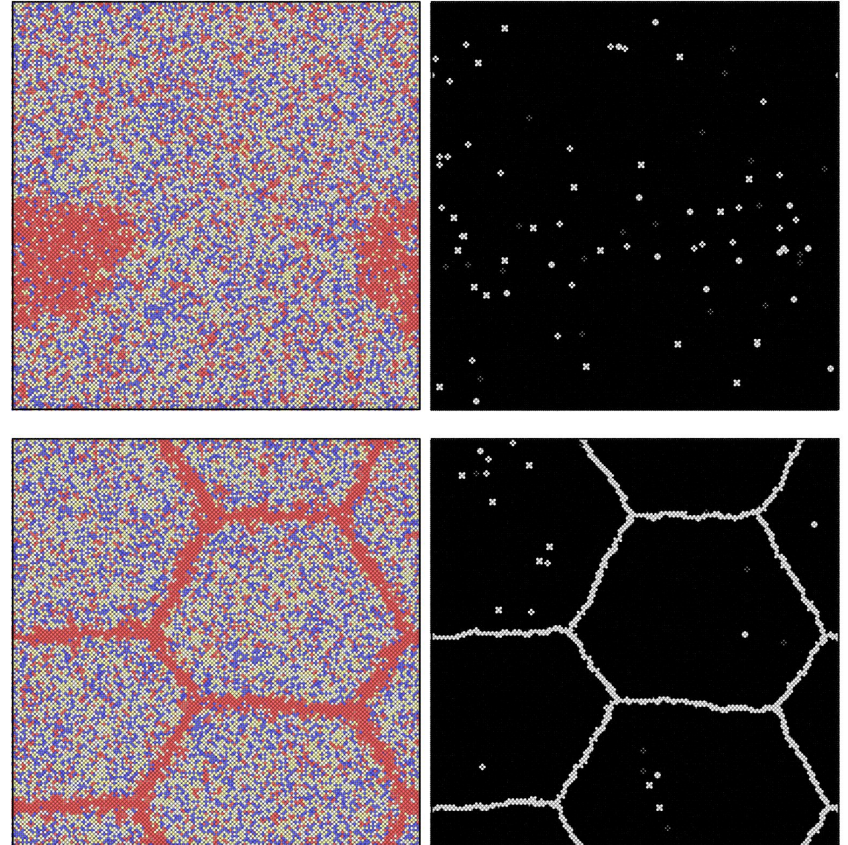
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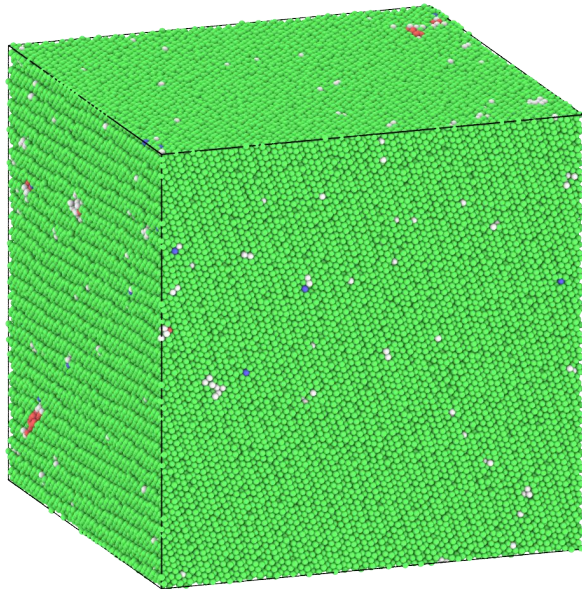


Segregation can strongly influence the final grain structure.

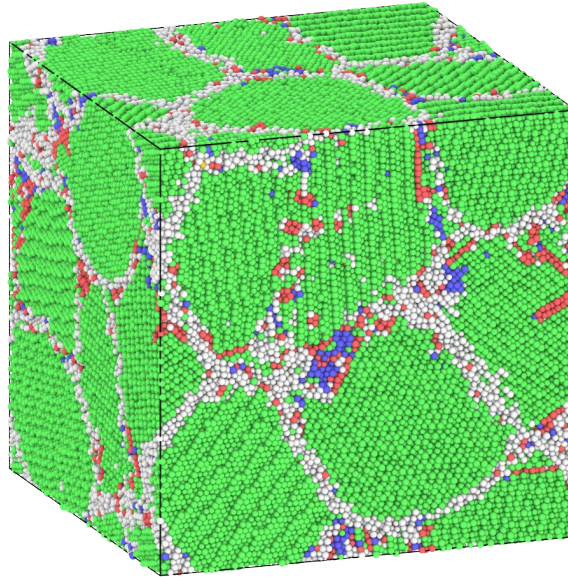


Conclusion

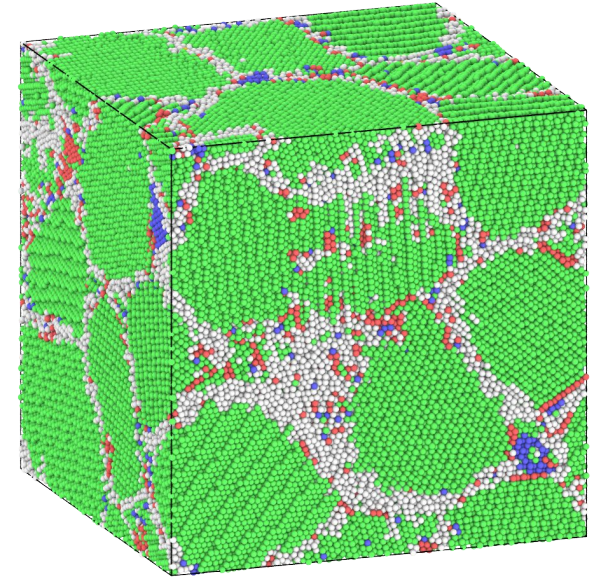
Cu



CuNiCoFe



Average Atom



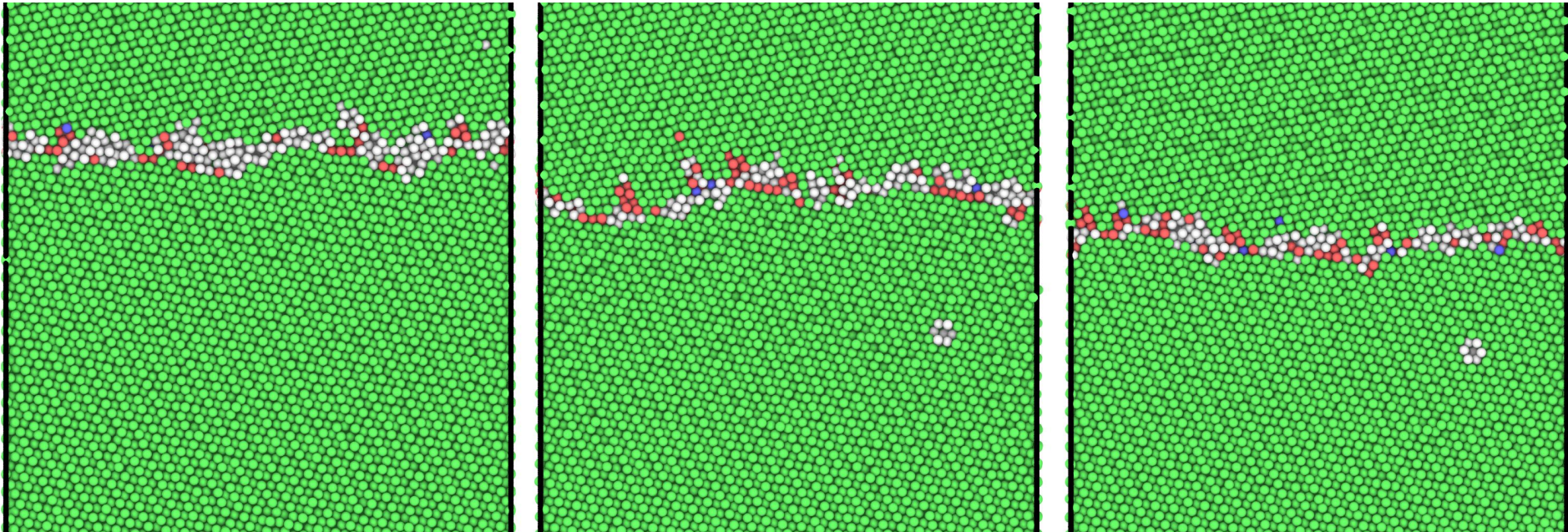
HEA shows strongly reduced grain growth.
It cannot be explained by local atomic disorder.

Conclusion

Cu

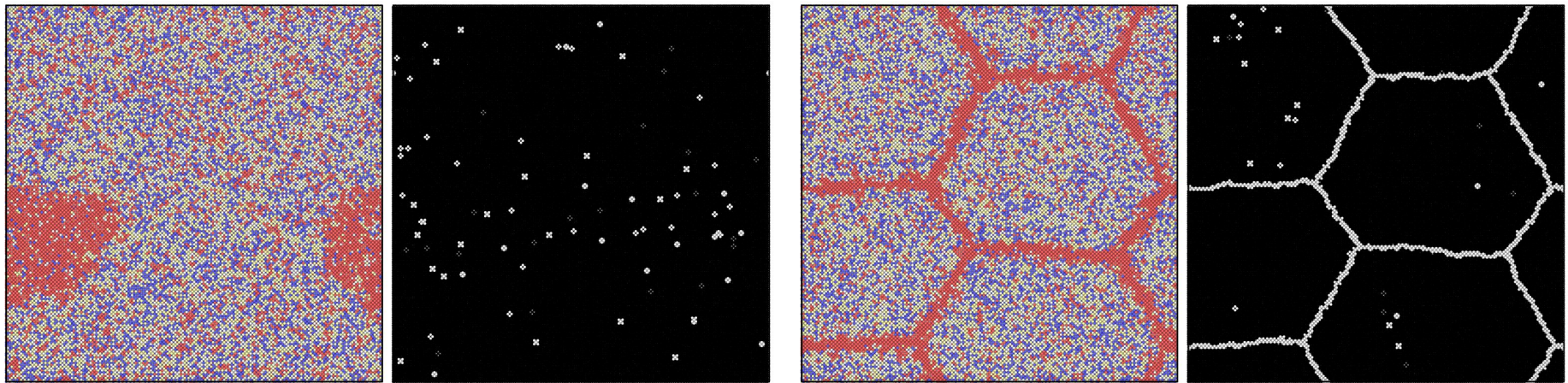
CuNiCoFe

Average Atom



HEA does not show significant signs of solute drag or Zener pinning.

Conclusion



Segregation effects need to be explored further.

Acknowledgement



SPP 2006: Legierungen mit komplexer Zusammensetzung - Hochentropielegierungen (CCA - HEA)



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Calculations for this research were conducted on the Lichtenberg high performance computer of the TU Darmstadt



Appendix Potentiale



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$$\phi(r) = \frac{A \exp[-\alpha(r/r_e - 1)]}{1 + (r/r_e - \kappa)^{20}} - \frac{B \exp[-\beta(r/r_e - 1)]}{1 + (r/r_e - \lambda)^{20}}, \quad (A3)$$

where r_e is the equilibrium spacing between nearest neighbors, A , B , α , and β are four adjustable parameters, and κ and λ are two additional parameters for the cutoff. The electron density function is taken with the same form as the attractive term in the pair potential with the same values of β and λ , i.e.,

144113-8

$$f(r) = \frac{f_e \exp[-\beta(r/r_e - 1)]}{1 + (r/r_e - \lambda)^{20}}, \quad (A4)$$

The pair potential between different species a and b is then constructed as

$$\phi^{ab}(r) = \frac{1}{2} \left[\frac{f^b(r)}{f^a(r)} \phi^{aa}(r) + \frac{f^a(r)}{f^b(r)} \phi^{bb}(r) \right]. \quad (A5)$$

Embedding energy functions that work well over a wide range of electron density require that three equations be used

MISFIT-ENERGY-INCREASING DISLOCATIONS IN ...

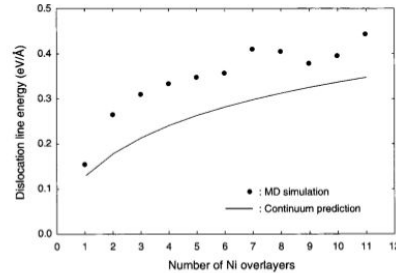
to separately fit three different electron density ranges. For a smooth variation of the embedding energy, these equations are required to match values and slopes at their junctions. These equations are

$$F(\rho) = \sum_{i=0}^3 F_{ni} \left(\frac{\rho}{\rho_n} - 1 \right)^i, \quad \rho < \rho_n, \quad \rho_n = 0.85\rho_e, \quad (A6)$$

$$F(\rho) = \sum_{i=0}^3 F_i \left(\frac{\rho}{\rho_e} - 1 \right)^i, \quad \rho_n \leq \rho < \rho_0, \quad \rho_0 = 1.15\rho_e, \quad (A7)$$

$$F(\rho) = F_e \left[1 - \ln \left(\frac{\rho}{\rho_s} \right)^\eta \right] \left(\frac{\rho}{\rho_s} \right)^\eta, \quad \rho_0 \leq \rho. \quad (A8)$$

PHYSICAL REVIEW B 69, 144113 (2004)



Zhou et al., Physical Review B, 69(144113), 2004

$$\langle E_0 \rangle = \sum_{i,X} c_X \langle F^X(\rho_i) \rangle + \frac{1}{2} \sum_{\substack{i,j \neq i \\ X,Y}} V_{ij}^{XY} c_X c_Y, \quad (3)$$

where $\langle s_i^X \rangle = c_X$. We then perform a Taylor expansion $\langle F^X(\rho_i) \rangle = \langle F^X(\bar{\rho}_i) \rangle + O(\rho_i - \bar{\rho}_i)^2$ around the average electron density $\bar{\rho}_i$, in which the first order term vanishes since $\langle (\rho_i - \bar{\rho}_i) \rangle = 0$. Neglecting second and higher-order terms [34], which is the *only* approximation, the average energy is

$$\langle E_0 \rangle = \sum_i F^A(\bar{\rho}_i) + \frac{1}{2} \sum_{i,j \neq i} V_{ij}^{AA}, \quad (4)$$

$$\text{with } F^A(\bar{\rho}_i) = \sum_X c_X F^X(\bar{\rho}_i),$$

$$V_{ij}^{AA} = \sum_{X,Y} c_X c_Y V_{ij}^{XY}, \quad \bar{\rho}_i = \sum_{j \neq i} \sum_X c_X \rho_{ij}^X, \quad (5)$$

Varvenne et al., Physical Review B, 93(104201), 2016

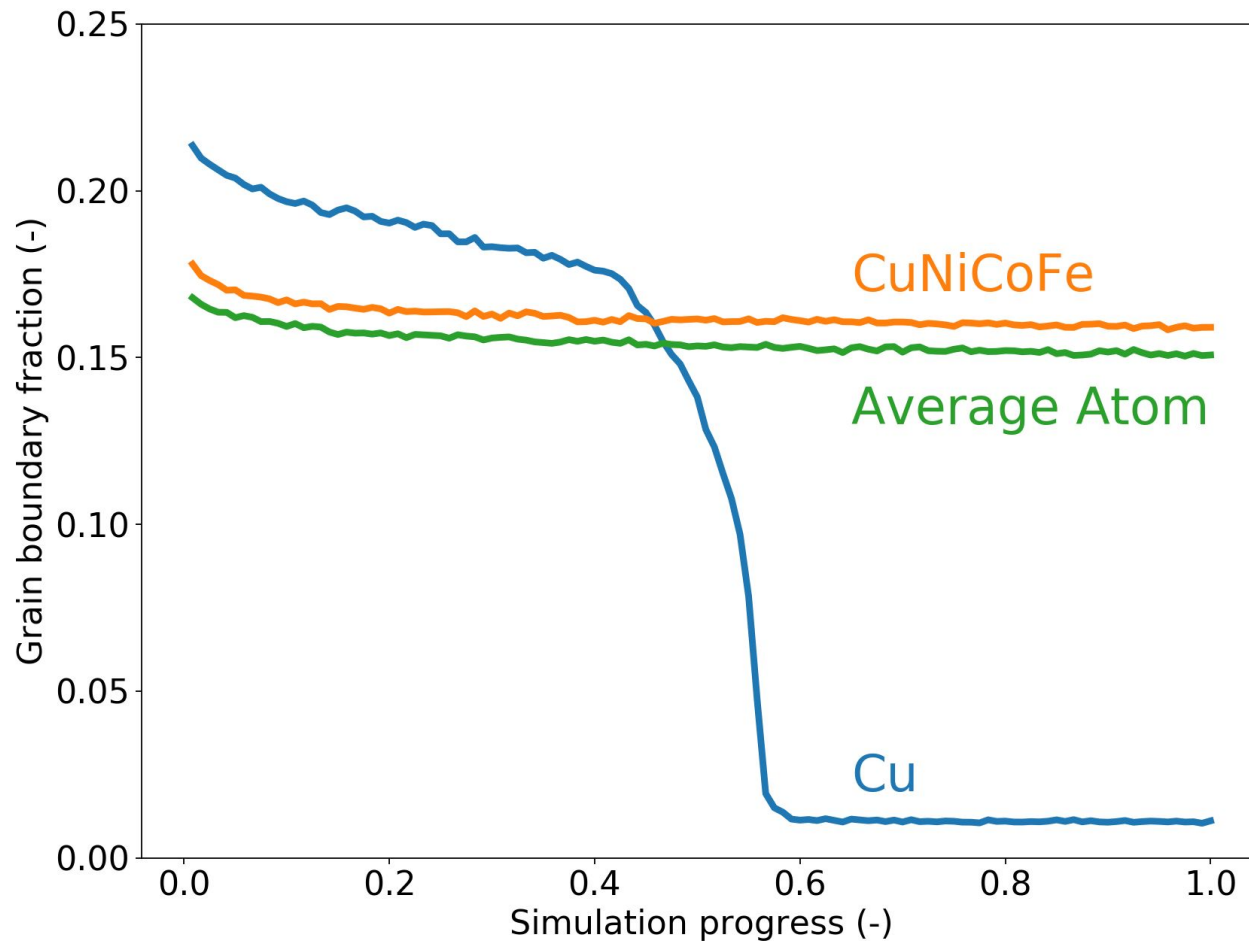
Appendix Average Atom

	CuNiCoFe	Average Atom
a_0 (Å)	3.572	3.570
E_{Coh} (eV)	-4.141	-4.137
C_{11} (GPa)	172	170
C_{44} (GPa)	100	101
C_{12} (GPa)	124	120
γ_{SF} (mJ/m ²)	10.6	26.9
γ_{USF} (mJ/m ²)	44.8	129.4

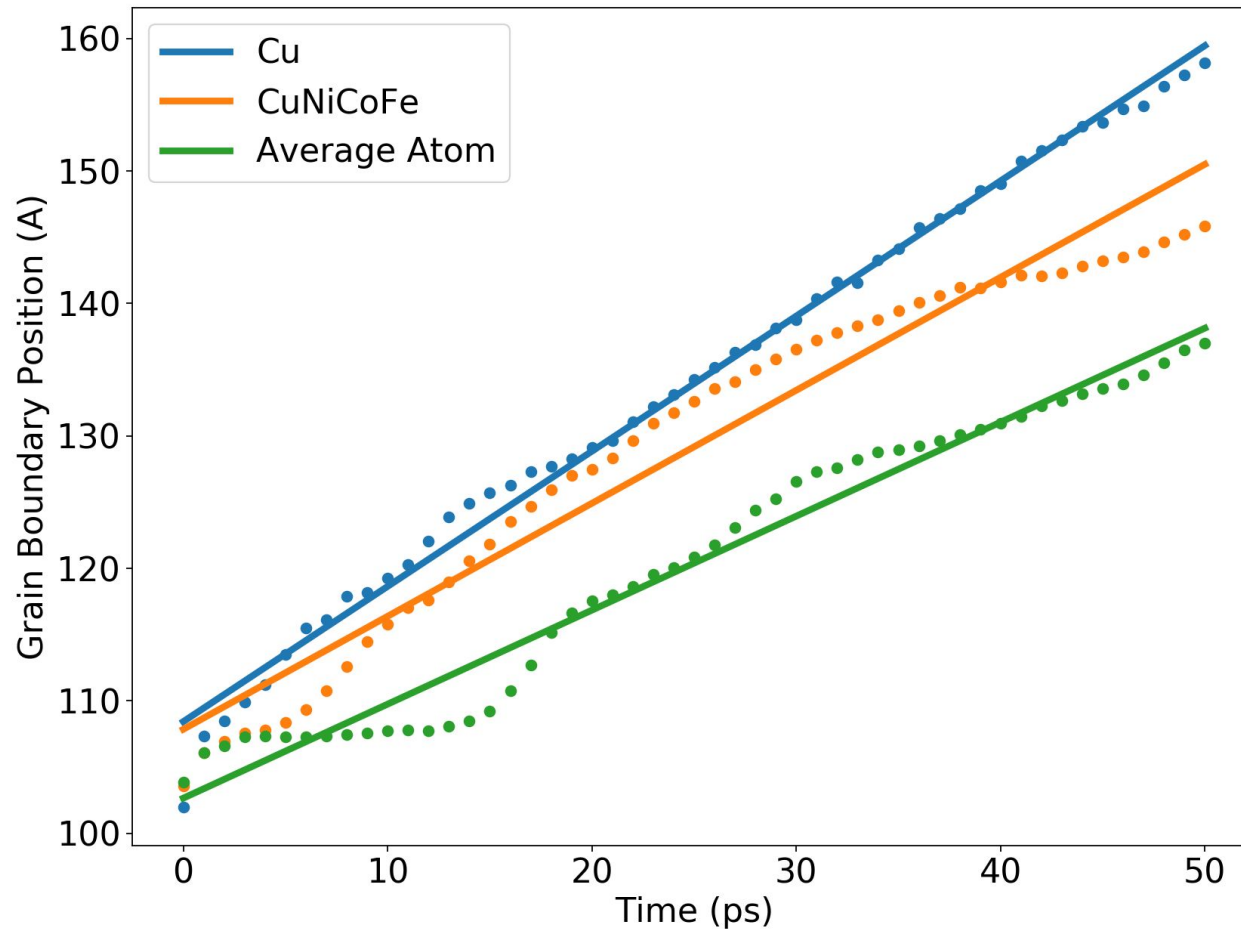
Appendix Nanocrystal



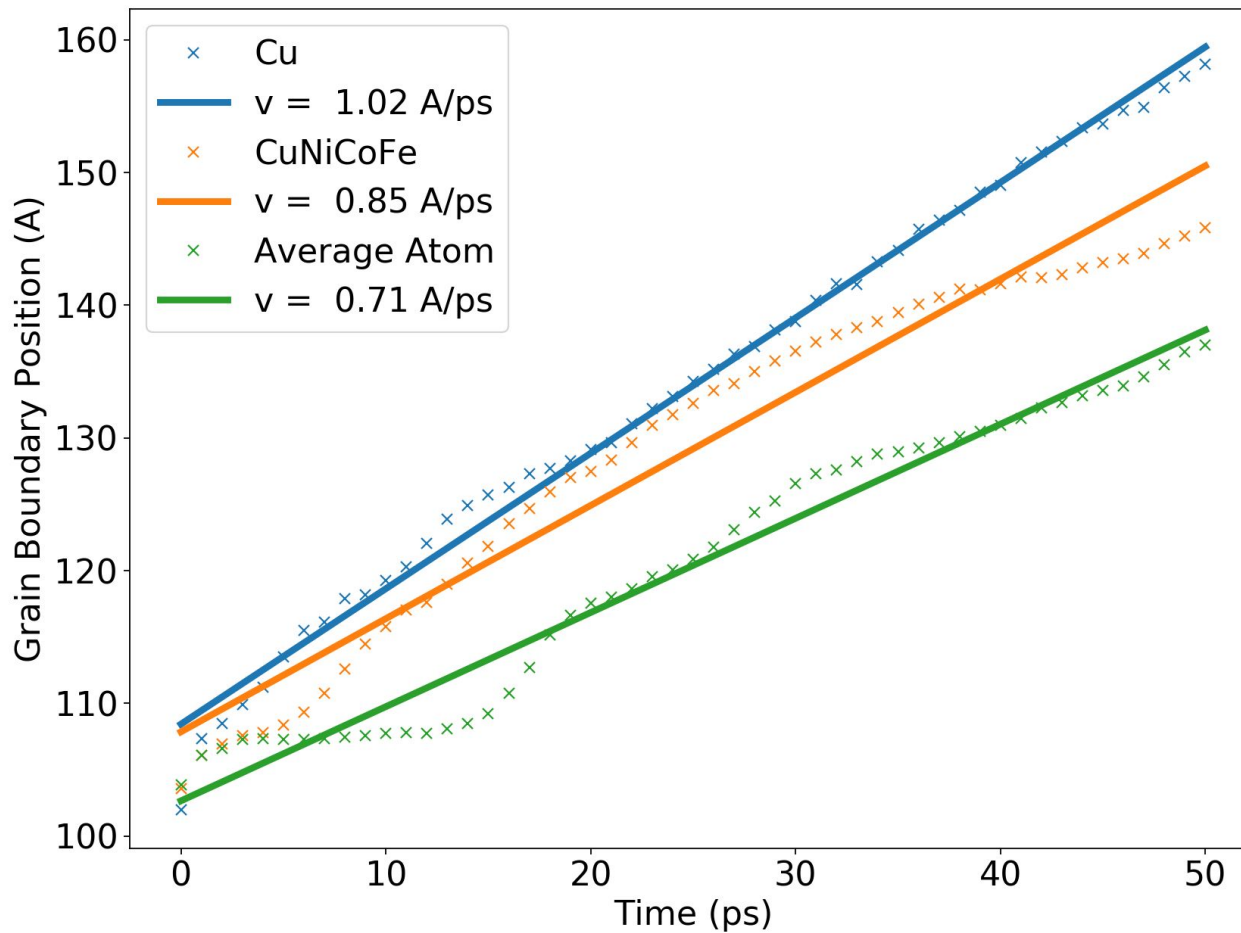
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Results



Appendix GB velocity



Appendix

Table 2.1: Bond energies in a polycrystalline binary alloy in full and simplified forms.

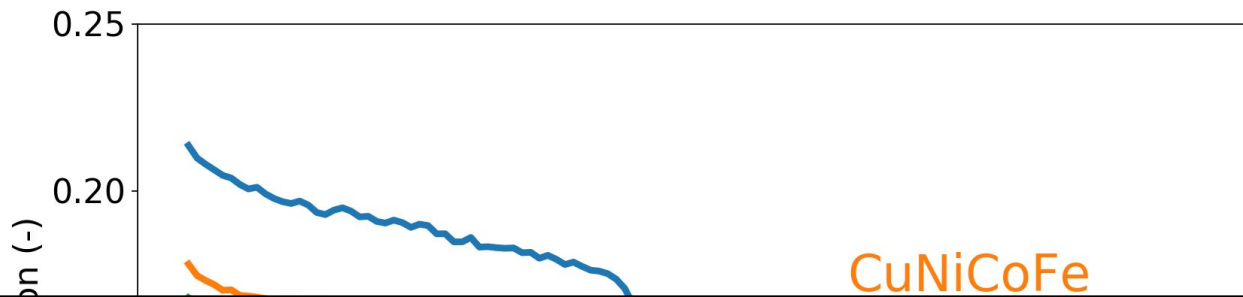
Bond energy	Full form	Simplified form
E_c^{AA}	E_c^{AA}	E_c^{AA}
E_c^{BB}	E_c^{BB}	E_c^{AA}
E_c^{AB}	$\left(\frac{E_c^{AA} + E_c^{BB}}{2}\right) + \omega_c$	$E_c^{AA} + \omega_c$
E_{gb}^{AA}	$E_c^{AA} + \frac{2\Omega^A\gamma_0^A}{zt}$	$E_c^{AA} + \frac{2\Omega^A\gamma_0^A}{zt}$
E_{gb}^{BB}	$E_c^{BB} + \frac{2\Omega^B\gamma_0^B}{zt}$	$E_c^{AA} + \frac{2\Omega^A\gamma_0^A}{zt}$
E_{gb}^{AB}	$\left(\frac{E_c^{AA} + E_c^{BB}}{2}\right) + \omega_{gb} + \frac{1}{zt}(\Omega^A\gamma_0^A + \Omega^B\gamma_0^B)$	$E_c^{AA} + \omega_{gb} + \frac{2\Omega^A\gamma_0^A}{zt}$

$$\Delta H^{\text{mix}} = z\omega_c X(1 - X),$$

$$\Delta H^{\text{seg}} = z \left[\omega_c - \frac{\omega_{gb}}{2} - \frac{1}{2zt}(\Omega^B\gamma_0^B - \Omega^A\gamma_0^A) \right]$$

T. Chookajorn PhD Thesis, Massachusetts Institute of Technology, 2013

Results



Local disorder is not the main contributor to the reduced grain growth.

